

Nano-engineering of aqueous polymer latex particles for film formation applications using multiblock copolymers

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Background: Multiblock copolymer latex particle film formation



Aguirre *et al.* M. Polymer Colloids: Current Challenges, Emerging Applications, and New Developments. *Macromolecules* **2023**, *56* (7), 2579-2607. Brito, E.L. and N. Ballard, Film formation of hard-core/soft-shell latex particles. Journal of Polymer Science, **2023**. 61(5): p. 410-421.



Background: Multiblock copolymer synthesis via RAFT polymerization



ADVANTAGES << Compartmentalization >>

- High degree of livingness
- Fast polymerization rate
- Applicable to low *k*p monomers
- Environmentally friendly (water as media)

APPLICATIONS

- Nanomedicine
- Materials science
- Latexes films
- Many more



Multiblock copolymer films and mechanical properties



Thompson S.W. et al. Macromolecules, 2023. 56(23): p. 9711-9724.



1.5

90 nm

132 nm

Nanoreactor concept: Multiblock copolymers by RAFT emulsion polymerization





<< Compartmentalization >>

- ✤ High polymerization rate
- ✤ Low initiator-derived radical concentration
- \clubsuit Low k_p monomers





Methodology: Synthesis of amphiphilic macroRAFT agent





MacroRAFT optimization

Effect of hydrophobic chain length





Log M (g mol⁻¹)



Seed latex synthesis (nanoreactors): RAFT emulsion polymerization

PBA seed DP=200



Seed *D* vs macroRAFT hydrophobic block DP













MacroRAFT optimization

Effect of hydrophilic chain length







Seed latex synthesis (nanoreactors): RAFT emulsion polymerization N PBA seed DP=200 °C₁₂H₂₅ HO 01 OH Seed *D* vs macroRAFT hydrophobic block DP 1.6 1.5 1.4 1.3 • Æ 1.2 • 1.1 1 80 100 20 40 60 120 0 Hydrophilic Block DP

High viscosity by increasing the hydrophobic DP

Styrene hexablock copolymer latex particle films



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Styrene hexablock copolymer latex particle films



Poor mechanical properties by increasing the styrene block numbers



Future work Sequential copolymerization **Microphase separation** Ν 111 0 $C_{12}H_{25}$ HO 200 II S /₂₀₀ /₆₀₀ 0 07 n **OH** 0 Ν III C 0 S $C_{12}H_{25}$ НО 300 II S /200 /200 300 Ο 0~ Ο О OH O

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