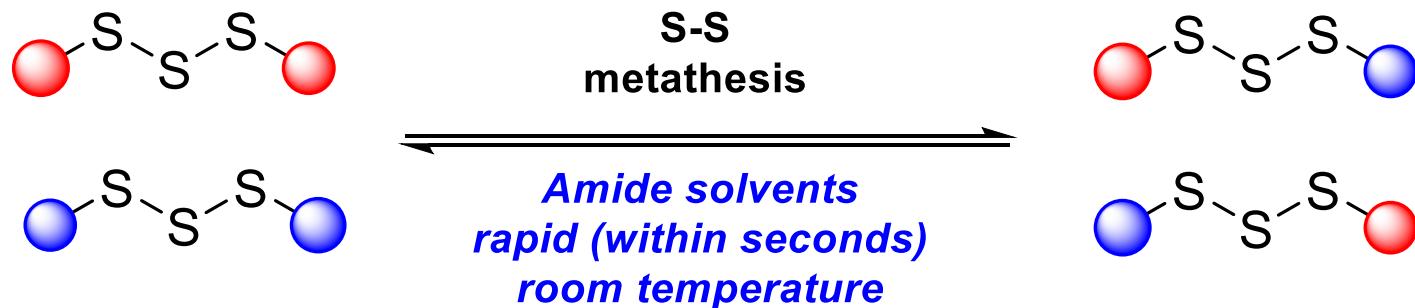


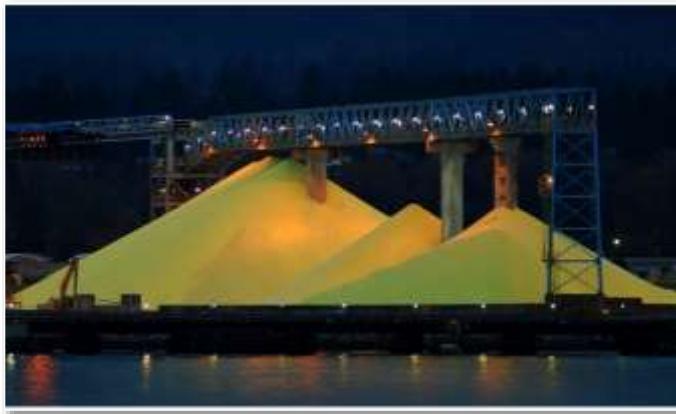
Amide Solvents Induce S-S Metathesis of Organic Trisulfides



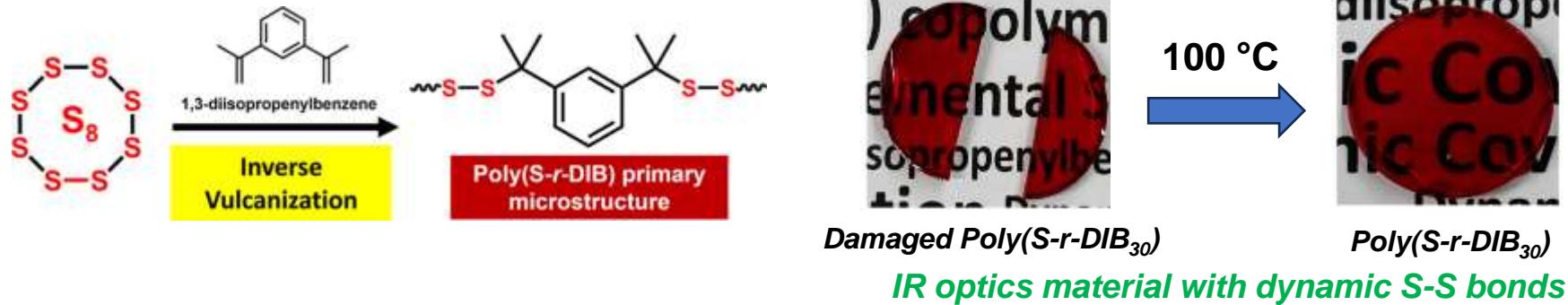
Alfrets D. Tikoalu
PhD Candidate

38th APS, 19 February 2024

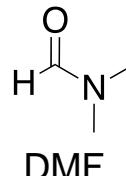
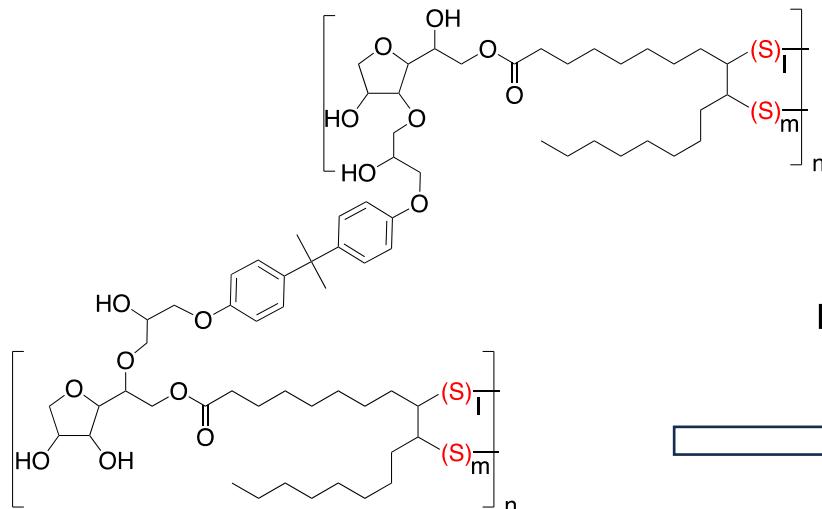
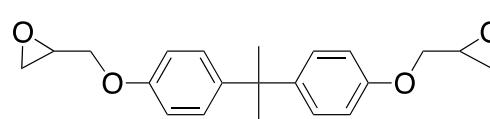
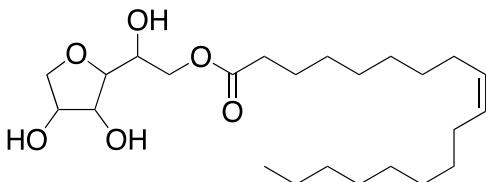
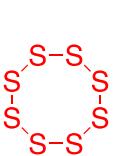
Background: Dynamic sulfur polymers for advanced materials



- Product of oil refining
- Abundant & inexpensive (~ 0.2 USD/kg)
- Megatons stockpile
- Can be transformed into sulfur polymers:
 - Material for IR optics
 - Sorbents for toxic metals
 - Composites
 - Repairable rubbers



Background: Sulfur polymer M_n changes in DMF



DMF

Chemical repair and recycling



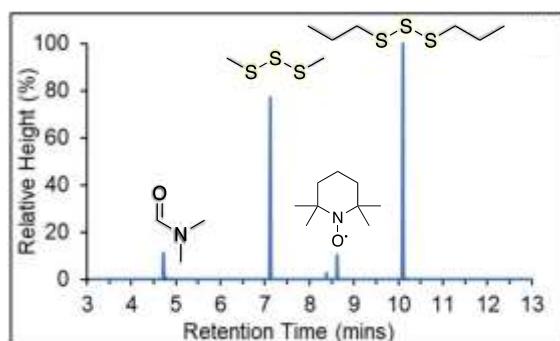
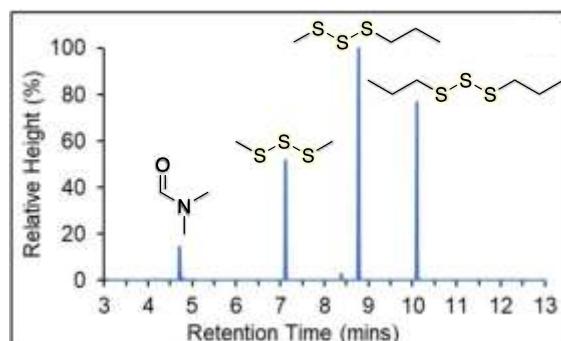
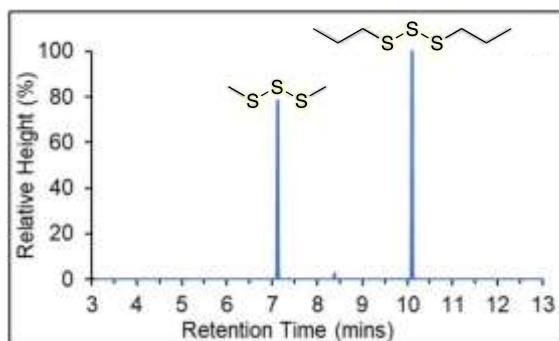
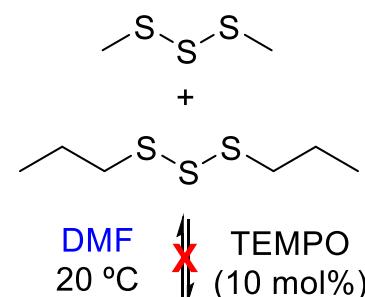
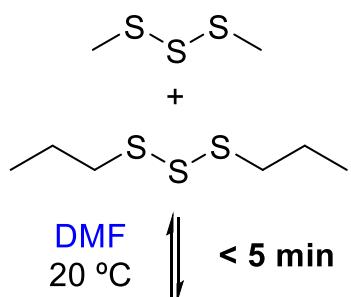
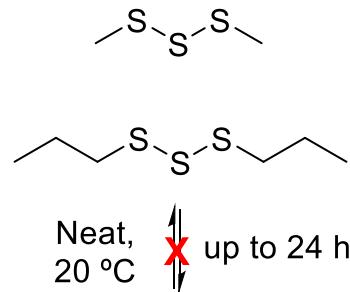
- Soluble (also in DMAc, NMP)
- Reduced M_n (by GPC)

Do these solvents actually break S-S bonds?

S-S bonds in trisulfides break reversibly in DMF



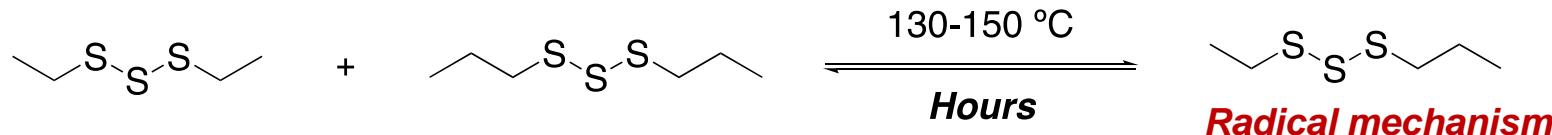
Sam Tonkin



S-S bond metathesis of organic trisulfides

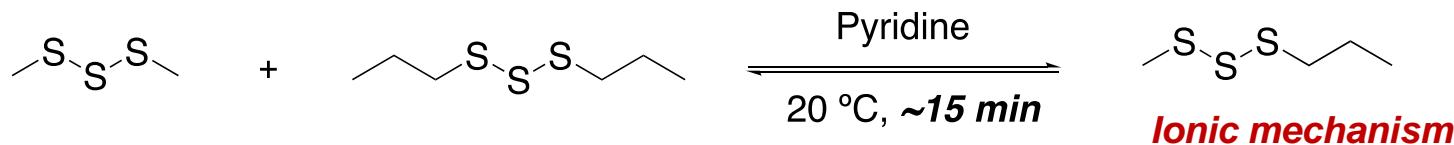


(Trivette and Coran, 1966)

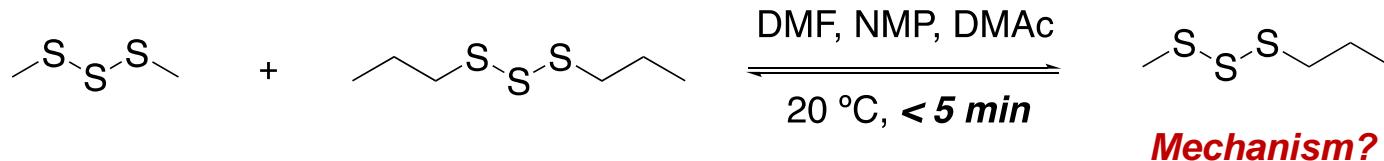


Sam Tonkin

(Tonkin, Chalker et al., 2020)

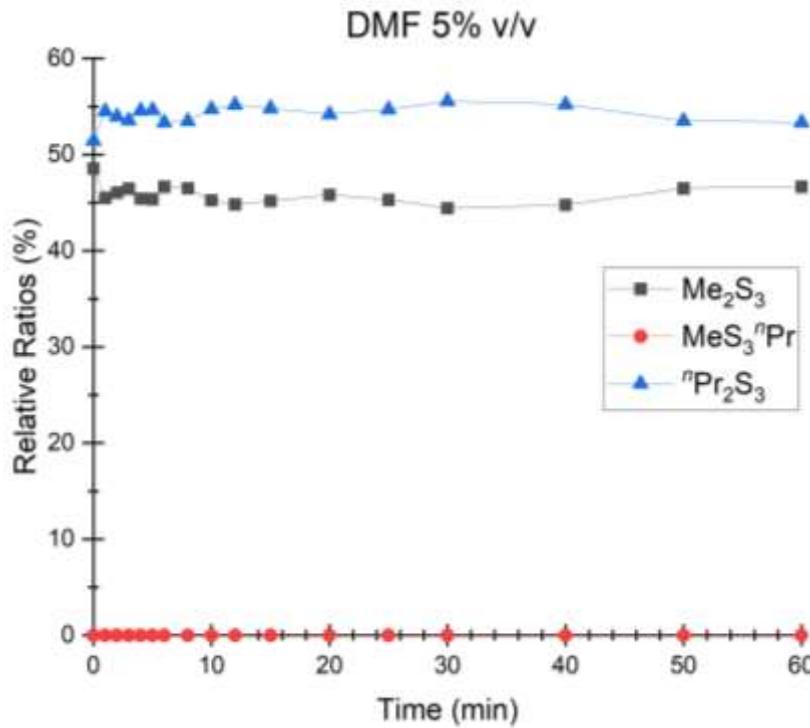
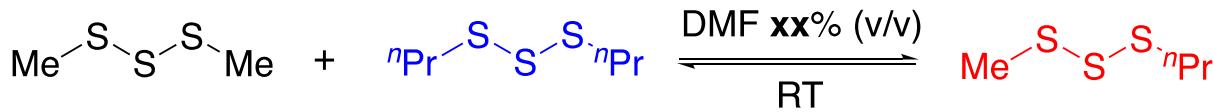


(Tonkin, Chalker, Hasell et al., 2022)

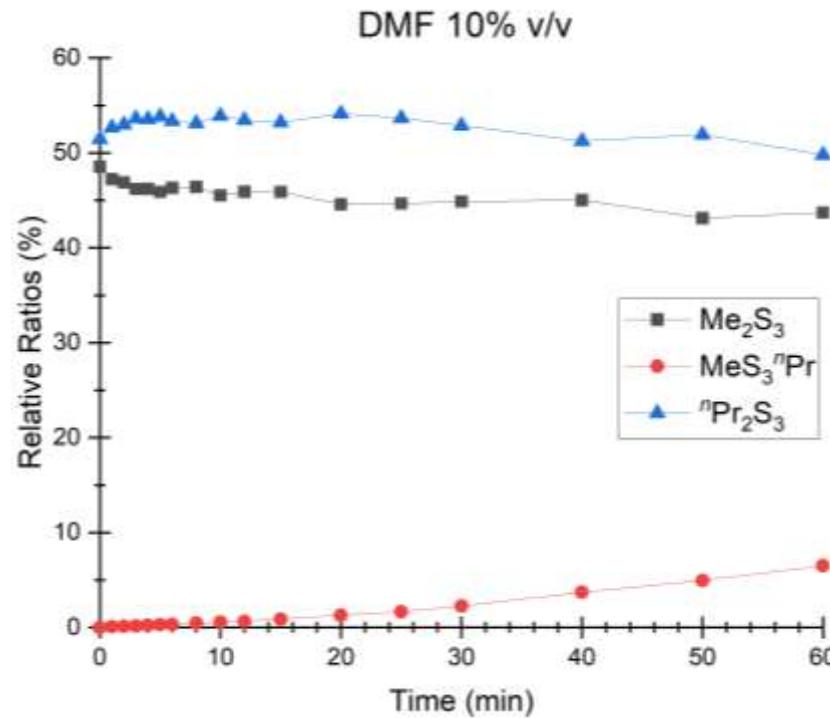
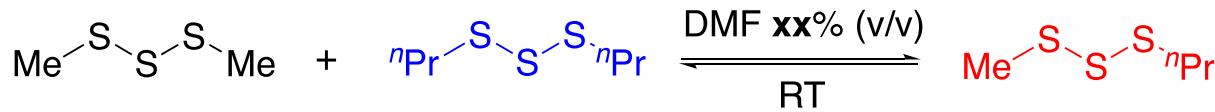


Trivette and Coran, *J. Org. Chem.* **1966**, 31, 100-104; Tonkin, Chalker et al. *Chem. Sci.*, **2020**, 11, 5537-5546;
Chalker, J. M., Hasell, T. et al. *Chem. Mater.* **2022**, 34, 3, 1167–1178;

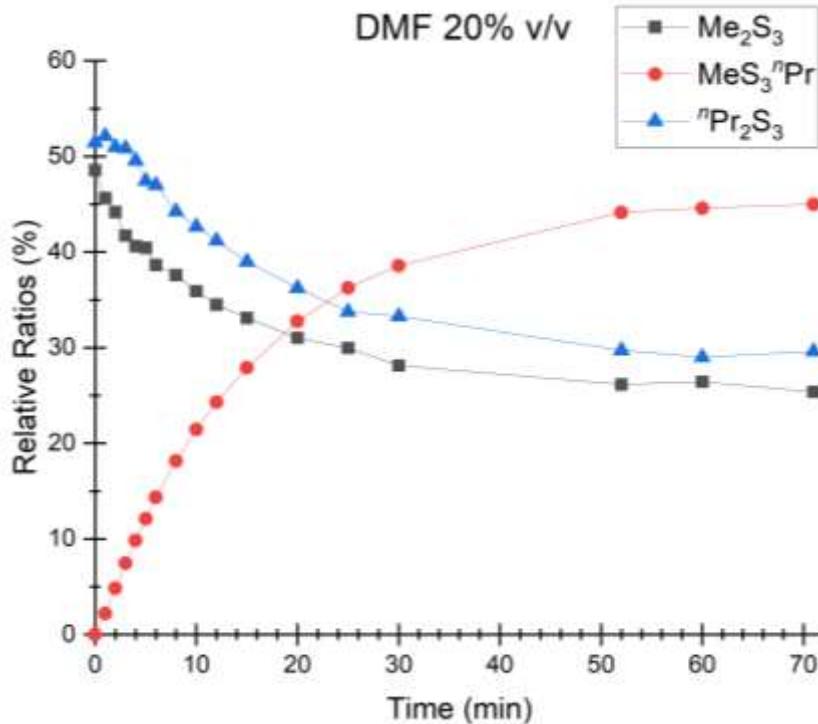
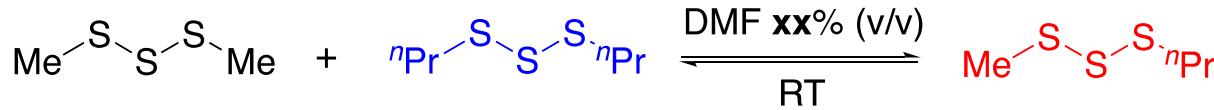
Solvent Investigation – amount of solvent



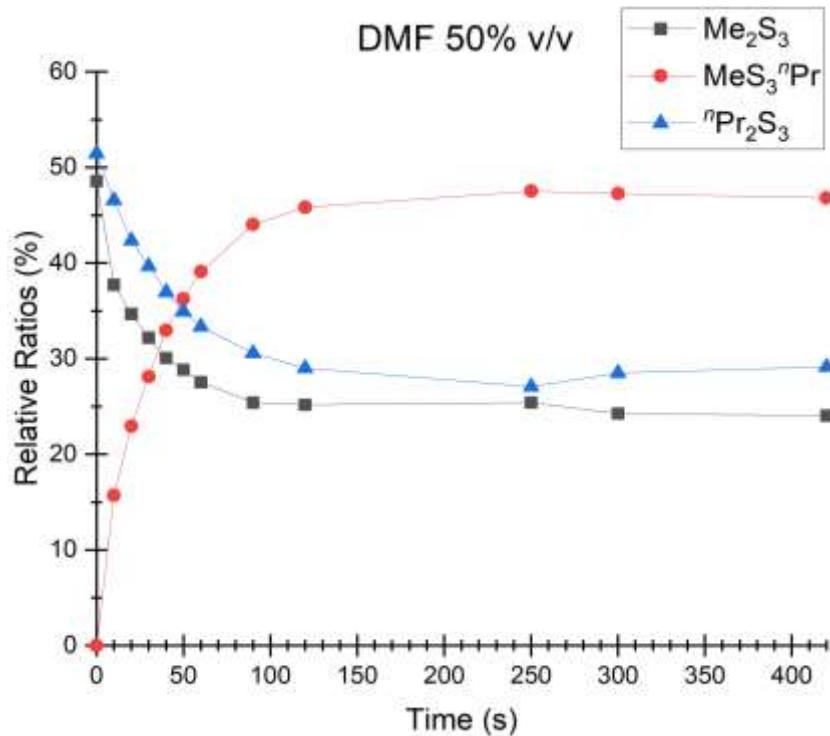
Solvent Investigation – amount of solvent



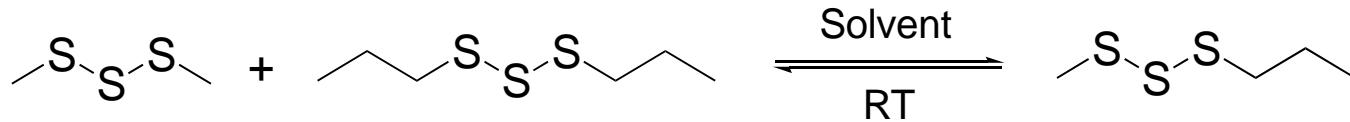
Solvent Investigation – amount of solvent



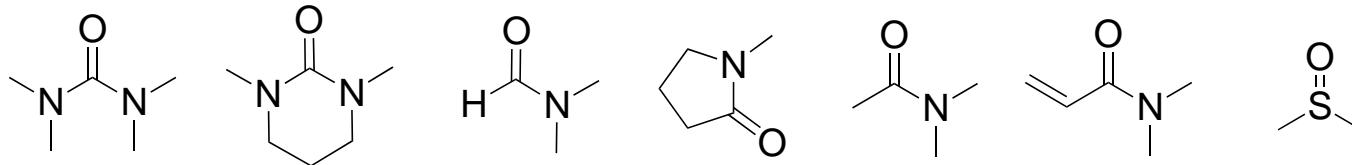
Solvent Investigation – amount of solvent



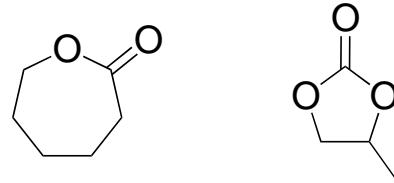
What solvents promote S-S metathesis in trisulfides?



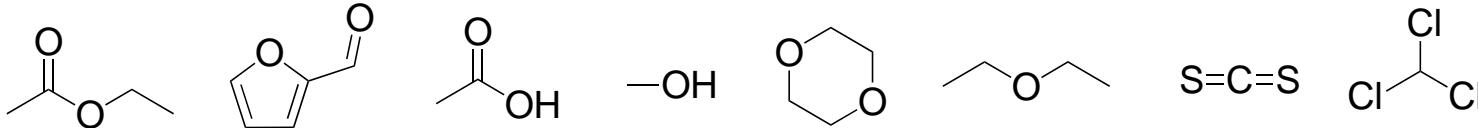
Metathesis within seconds:



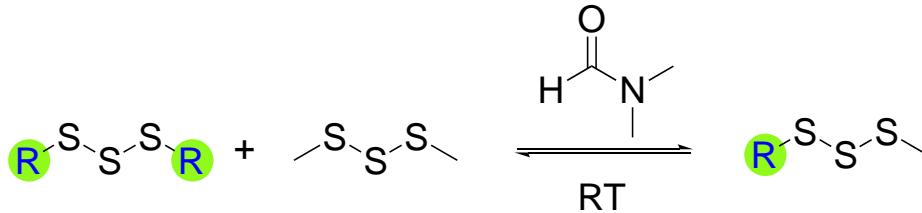
Metathesis within hours:



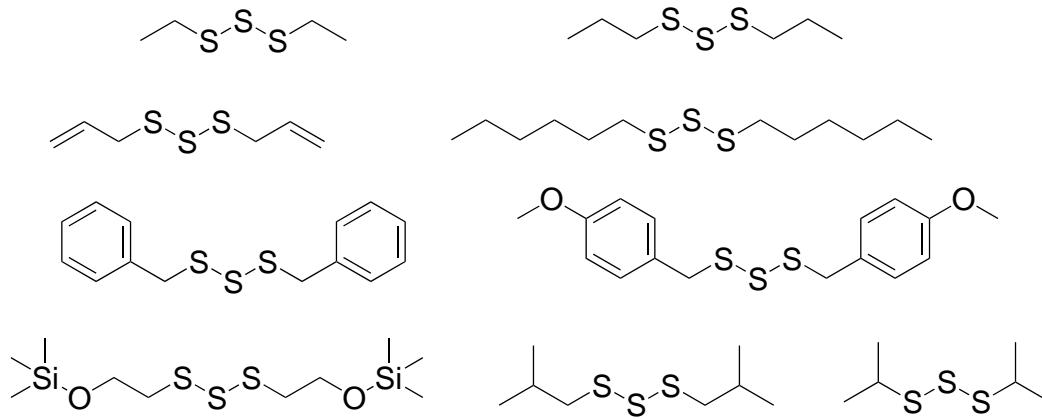
No reaction:



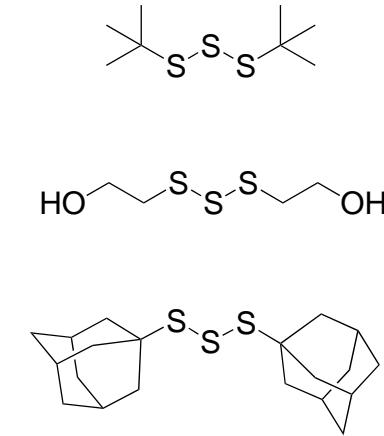
S-S metathesis – substrate scope



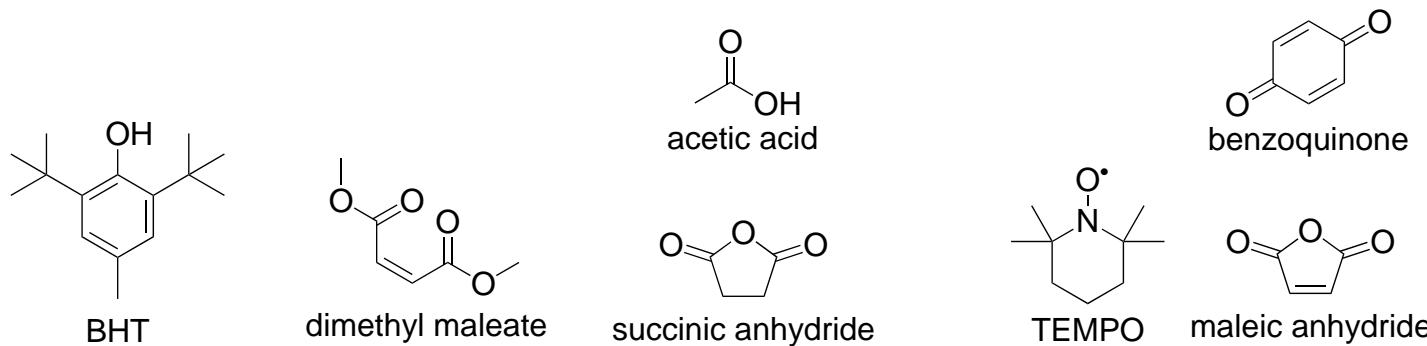
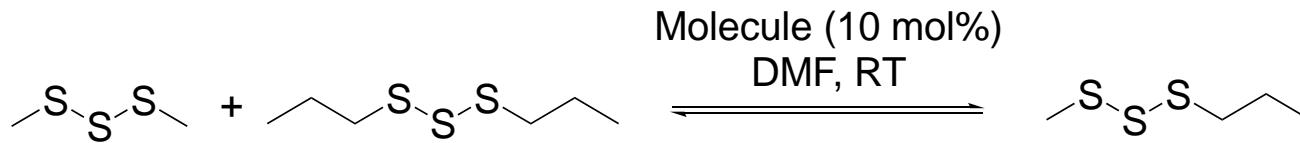
Rapid crossover in seconds



No crossover



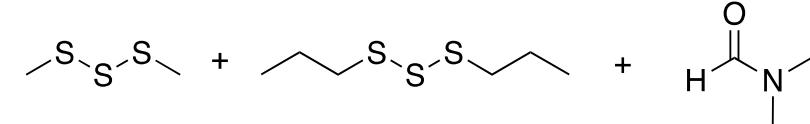
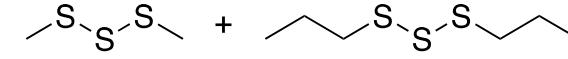
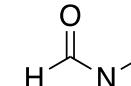
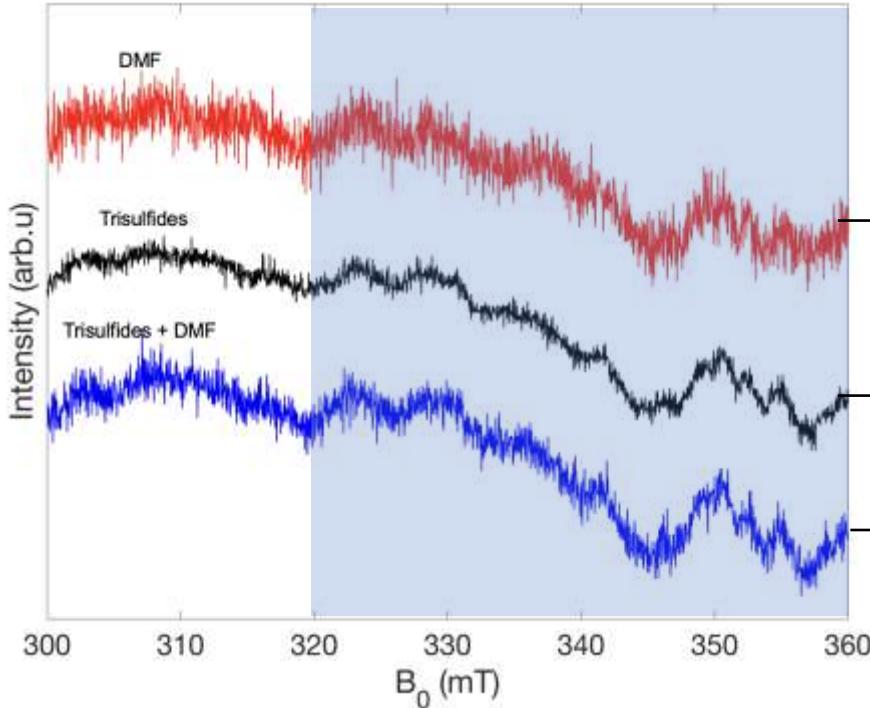
S-S metathesis is inhibited by electrophiles, acid, and oxidants



No inhibition

complete inhibition

EPR (Electron Paramagnetic Resonance) spectroscopy – no evidence for radical species

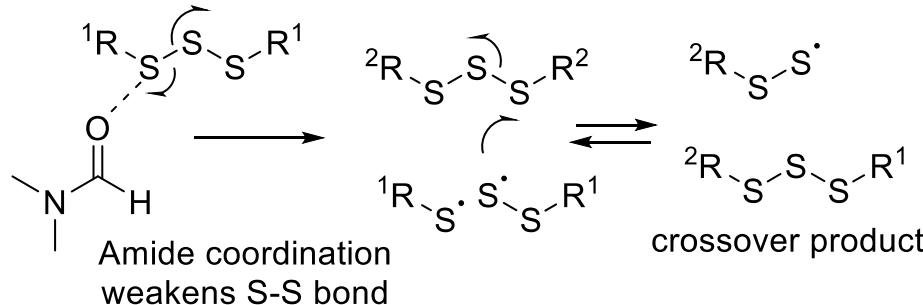


Result:
No S-centered radical signal was observed.

A/Prof. Jeffrey Harmer
(The University of Queensland)

Proposed mechanisms for the S-S metathesis

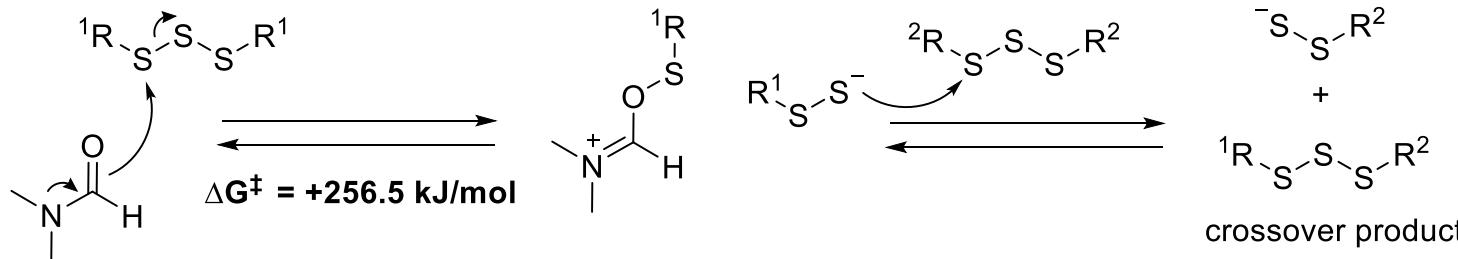
1. Radical mechanism



**No evidence
by EPR**



2. Ionic mechanism

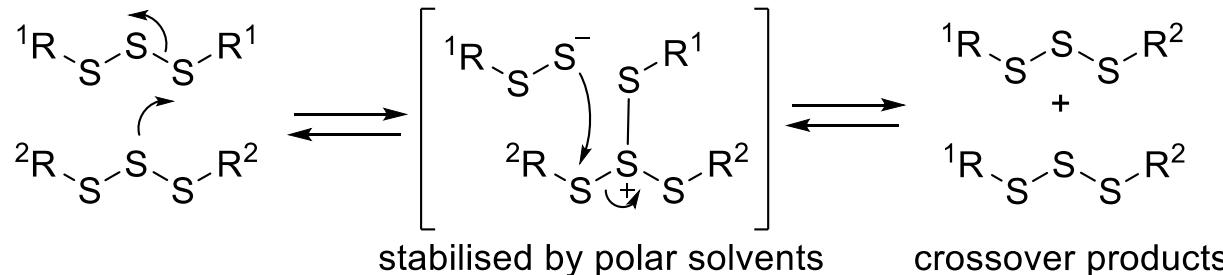


**Energy barrier
too high**

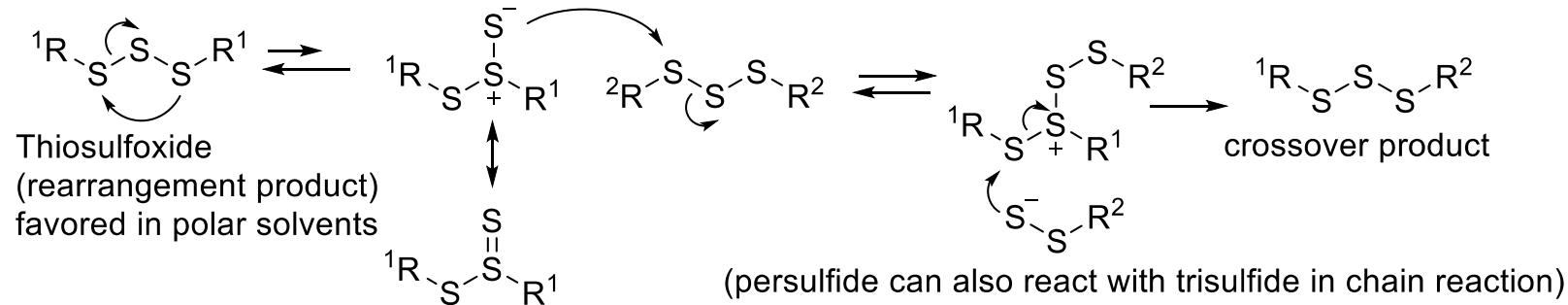


Proposed mechanisms for S-S metathesis

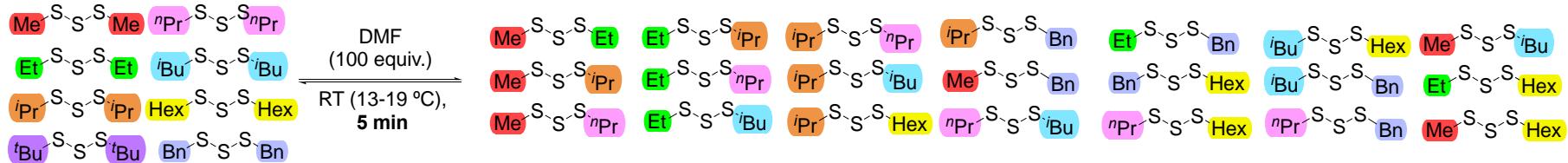
3. Ionic mechanism



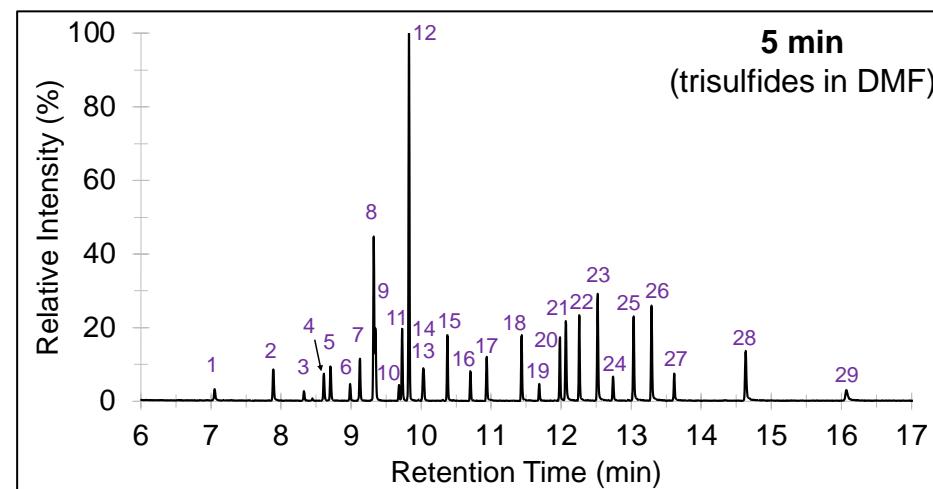
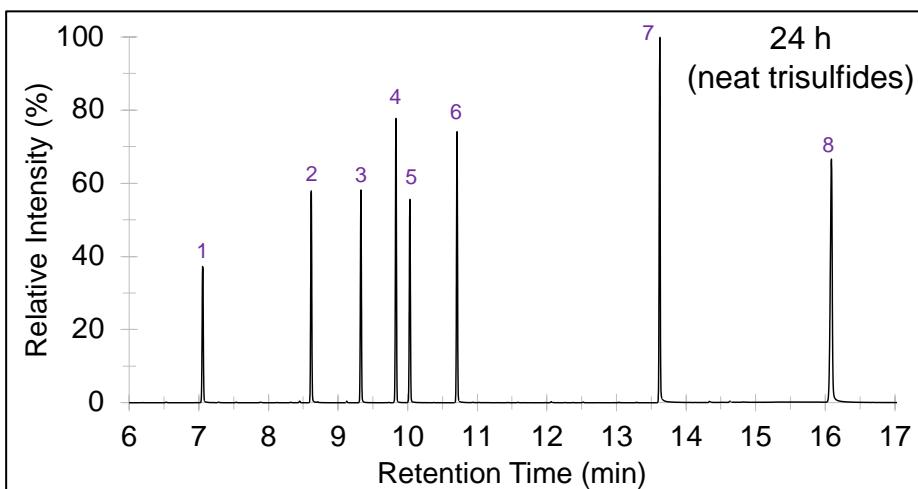
4. Ionic mechanism



Application – Dynamic Combinatorial Library



Rapid formation of dynamic library
8 to 29 trisulfides in < 5 minutes



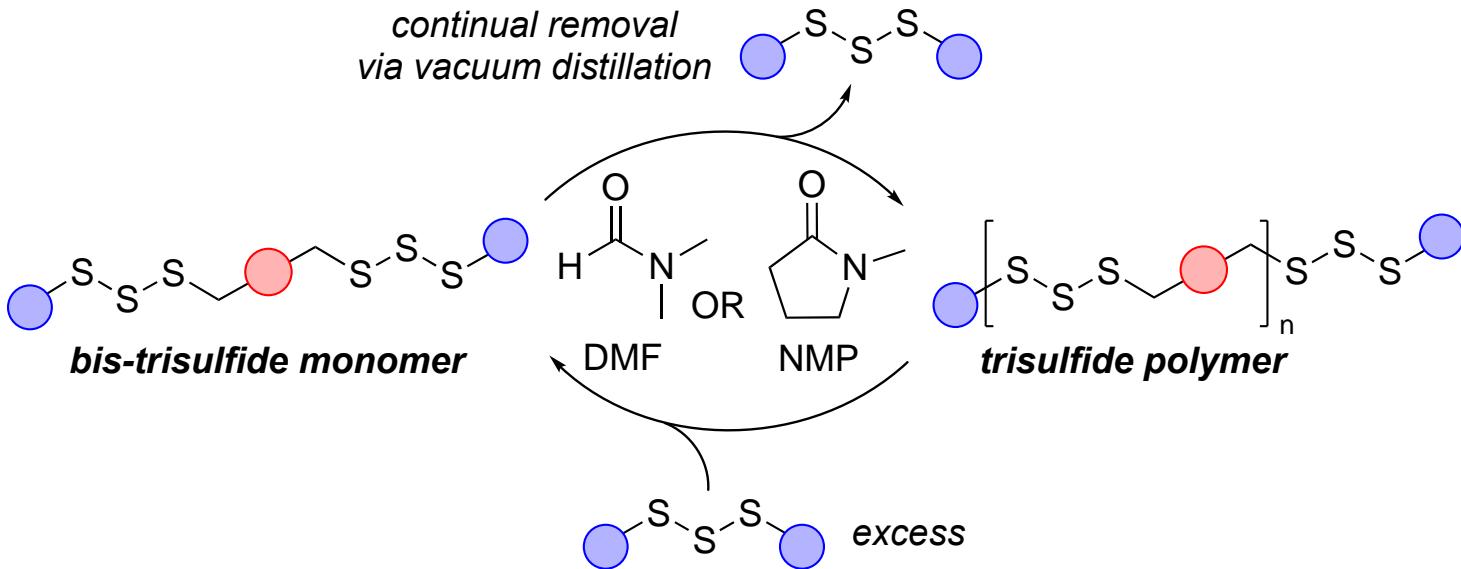
Application to recyclable polymers



Dr. Harshal D. Patel
Tasman 1, Tuesday
4:25-4.40 PM



James N. Smith
Poster Session



Provisional patent AU2024900381. Priority date 16 Feb 2024.

Acknowledgments

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A/Prof. Martin R. Johnston

Dr. Harshal Patel

Dr. Witold Bloch

Prof. Michele Coote

A/Prof. Jeffrey Harmer (UQ)

Chalker Lab (past and present members)

Synthesis Super Group at Flinders University



Australian Government
Australian Research Council

