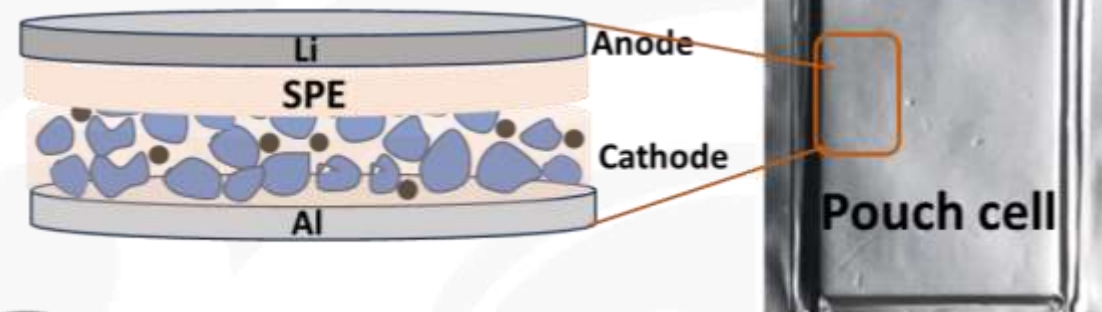


# Novel Solid-State Electrolytes for High Voltage Lithium and Sodium-Metal Batteries

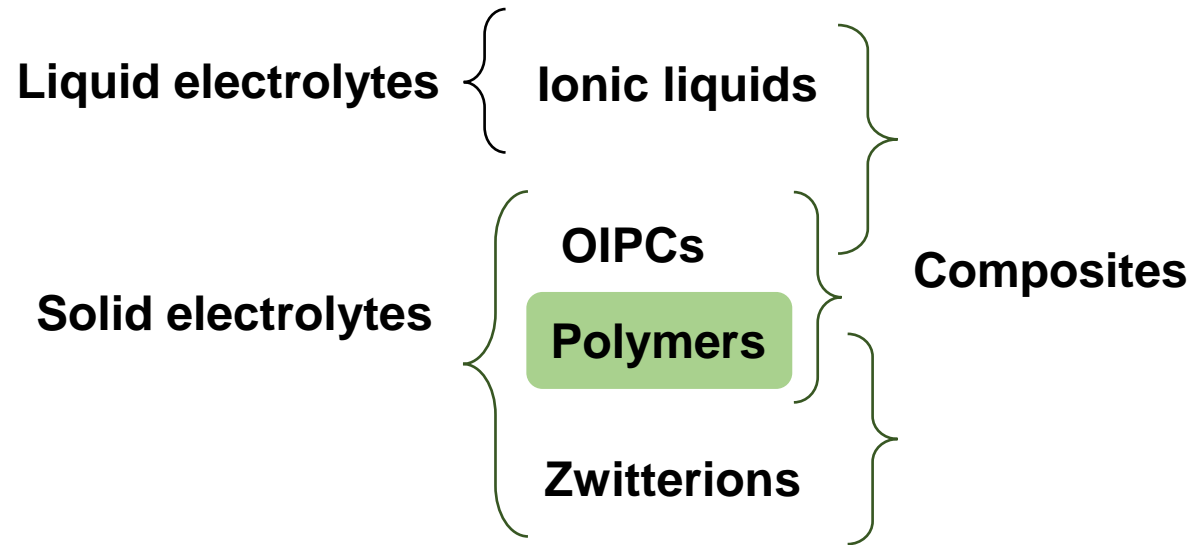
Faezeh Makhlooghiyazad

[f.makhlooghiyazad@deakin.edu.au](mailto:f.makhlooghiyazad@deakin.edu.au)

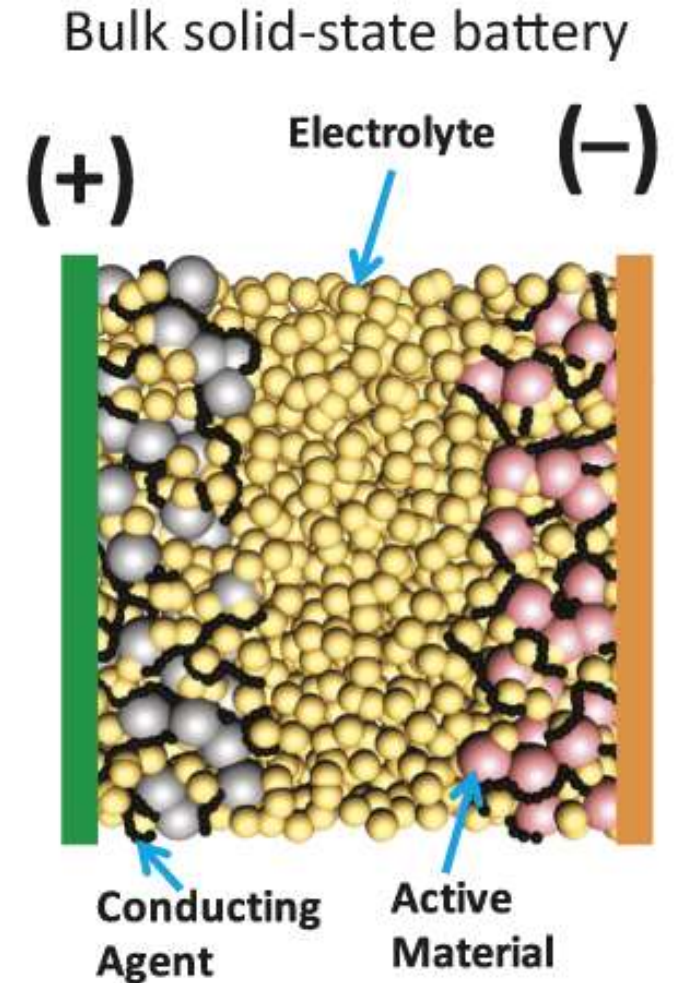
The Institute for Frontier Materials,  
Deakin University,  
Melbourne  
Australia

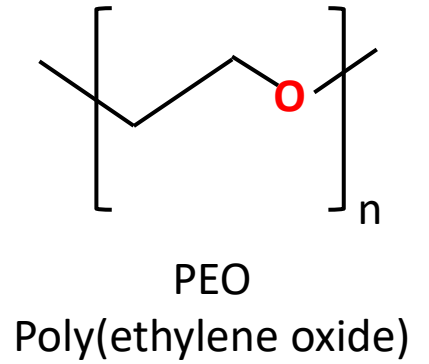


# Electrolytes for battery applications



- High ionic conductivity
- Chemical stability with electronically conductive additives
- Good mechanical properties
- Thermally & electrochemically stable
- Chemical stability with alkali metals





### Bluebus & Bluecar by Bolloré' [www.bluecar.fr](http://www.bluecar.fr)

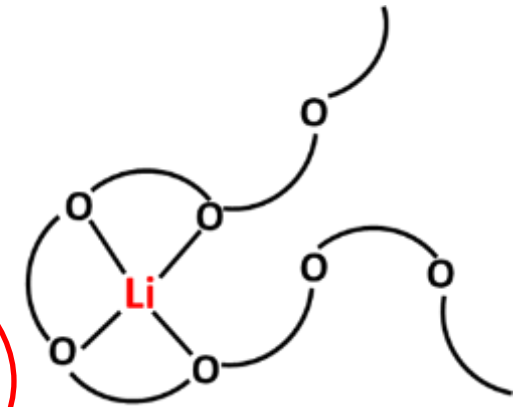
30 kWh, driving ranges of up to 250 km



- ✓ Chain flexibility
- ✓ Low T<sub>g</sub>
- ✓ great solubility for conductive Li salts



- × Poor anodic stability
- × High degree of crystallisation
- × Low ionic conductivity at RT
- × Low Li transference number



# Approaches to increase ionic conductivity

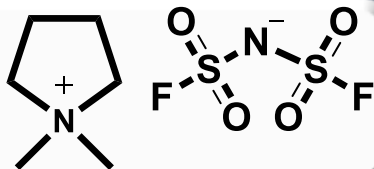
Adding plasticizers

Organic solvents

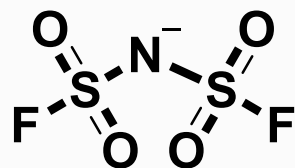
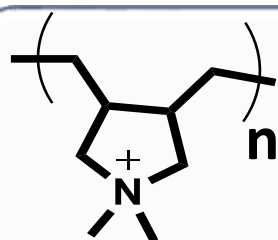
Flammable, Volatile

Ionic liquids

Safe



Polymerised ionic liquids

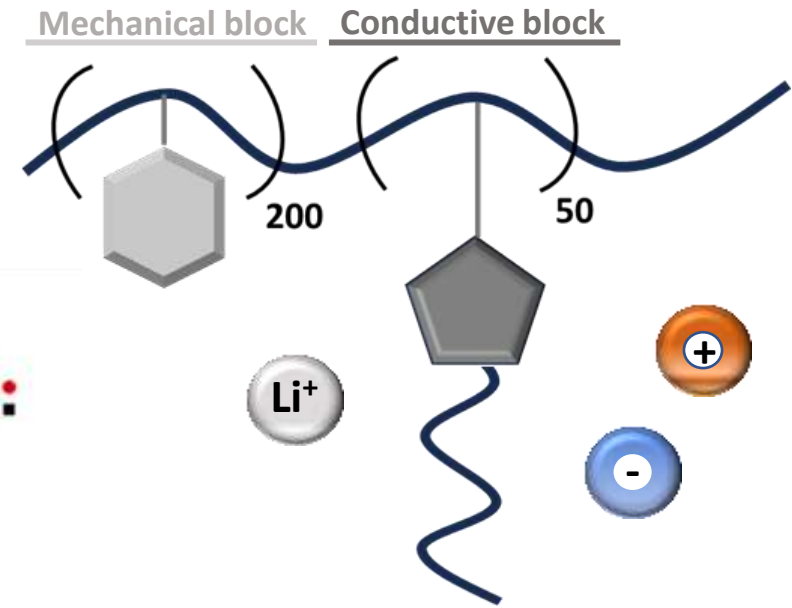
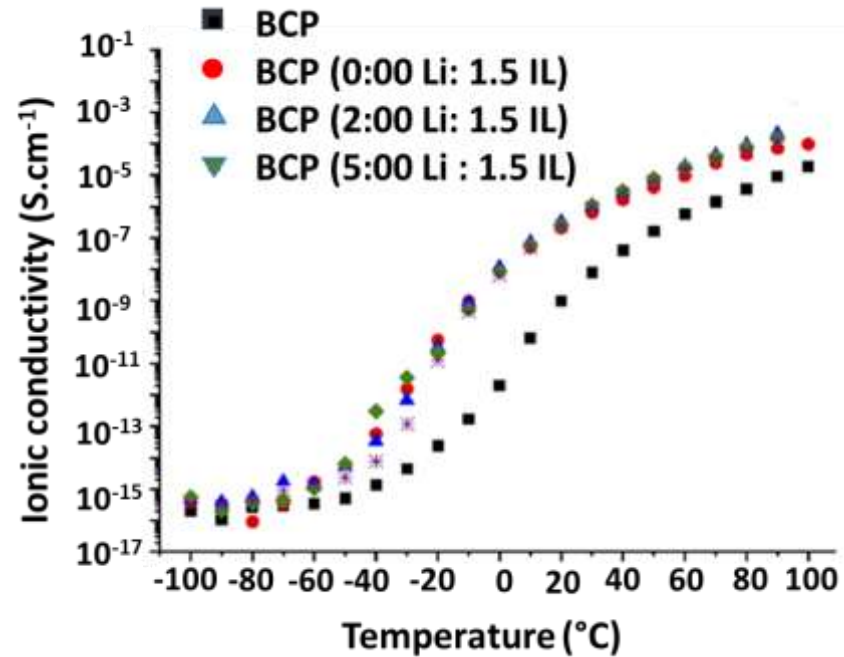
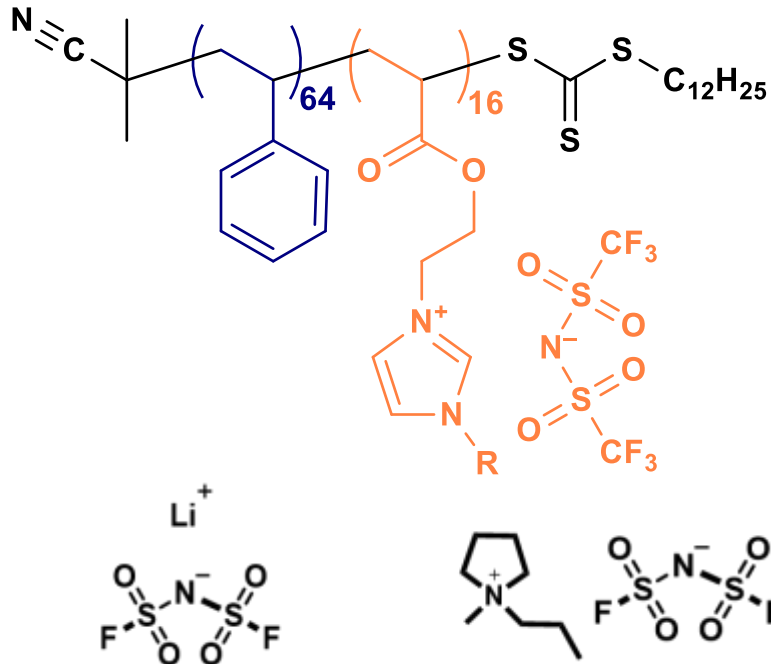


*Challenge*

Mechanical integrity & ionic conductivity



# Block Copolymer Electrolytes



Optimum composition

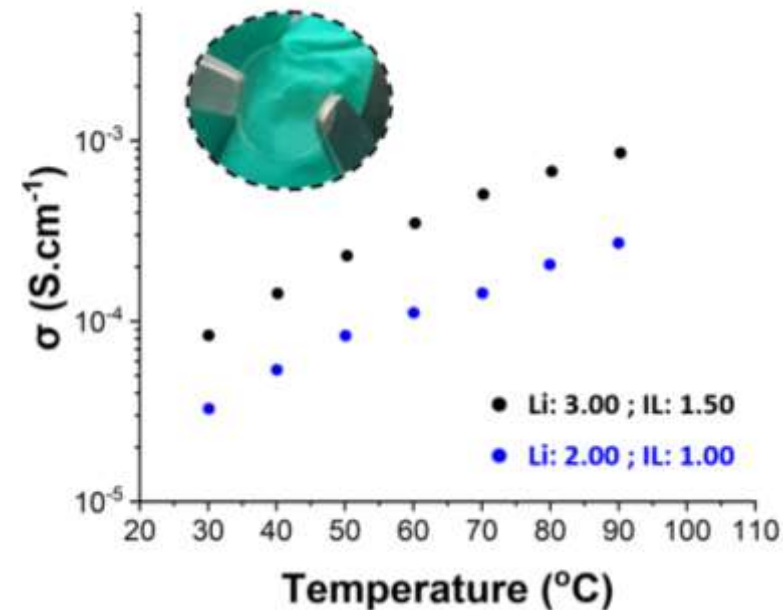
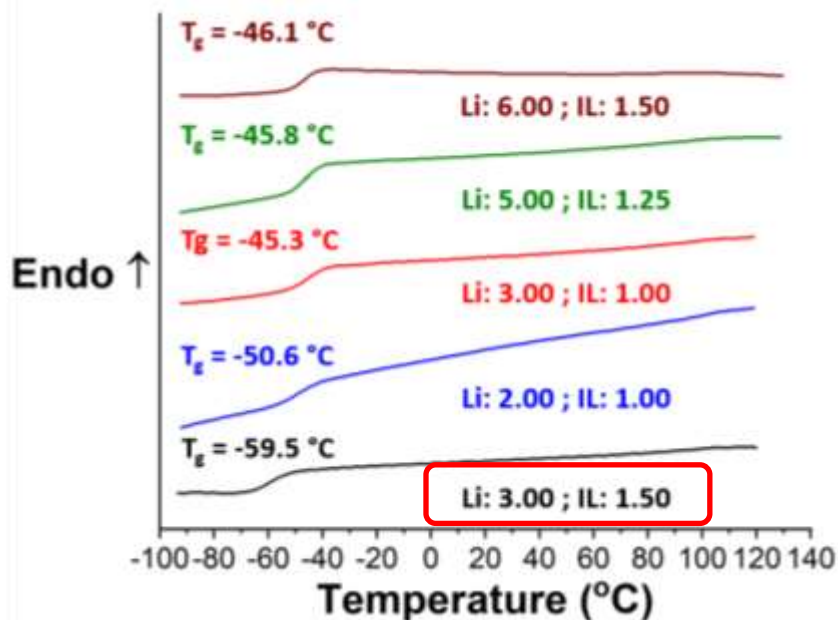
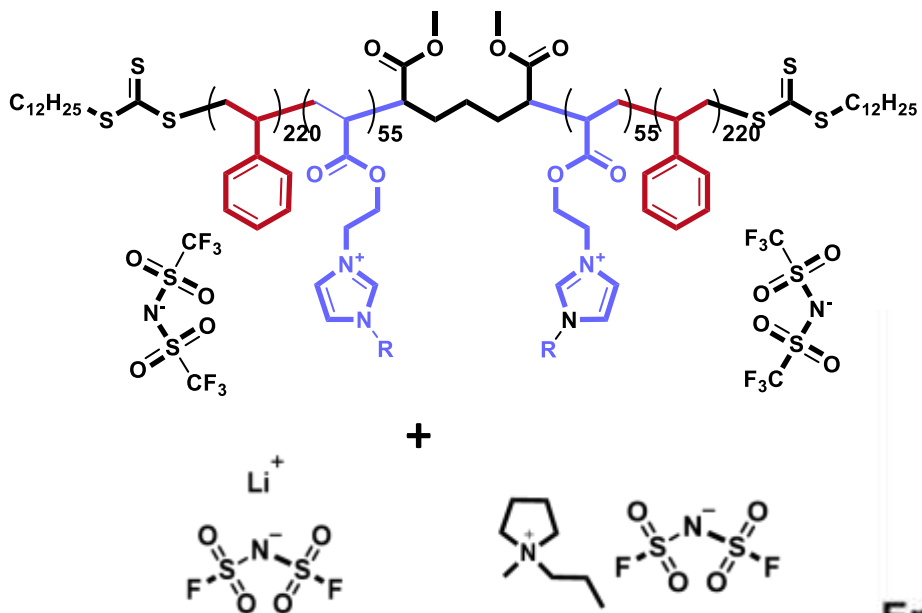
1:00 BCP : 2:00 LiFSI : 1.5 IL

Higher loadings of salt and IL

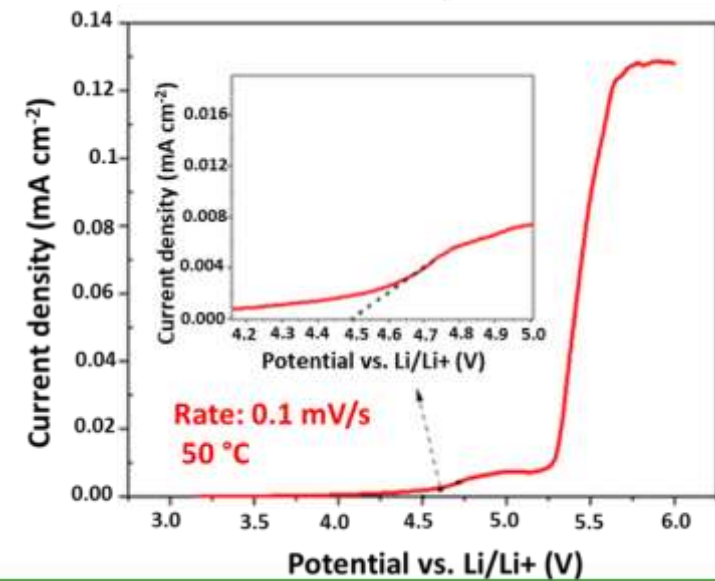
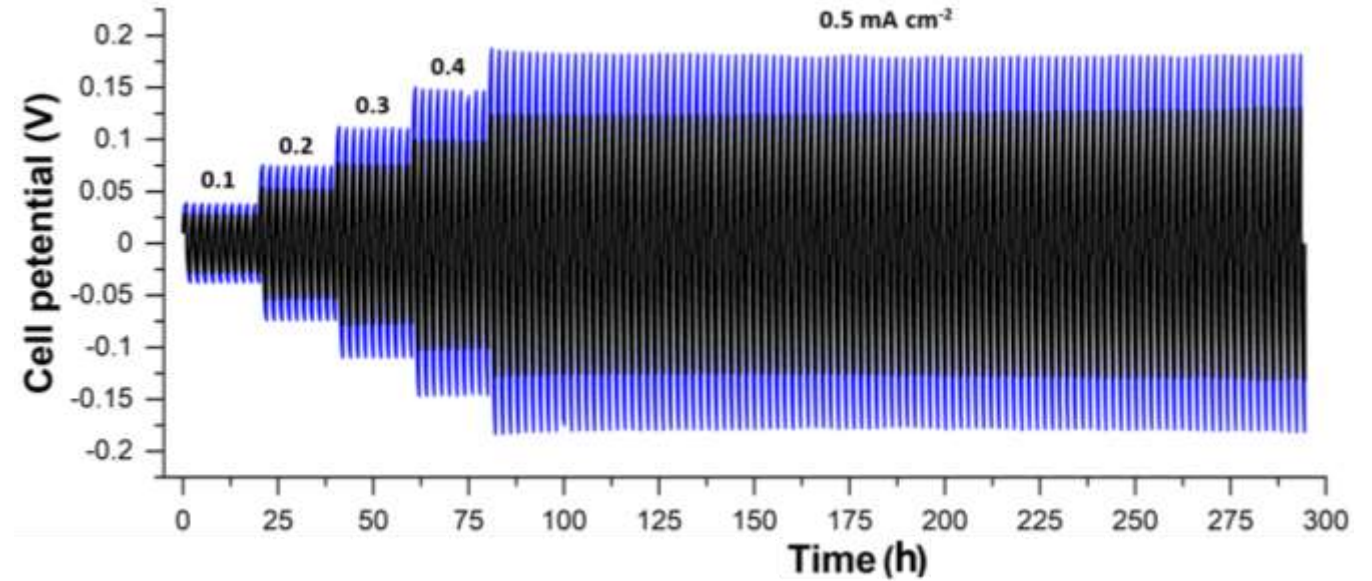
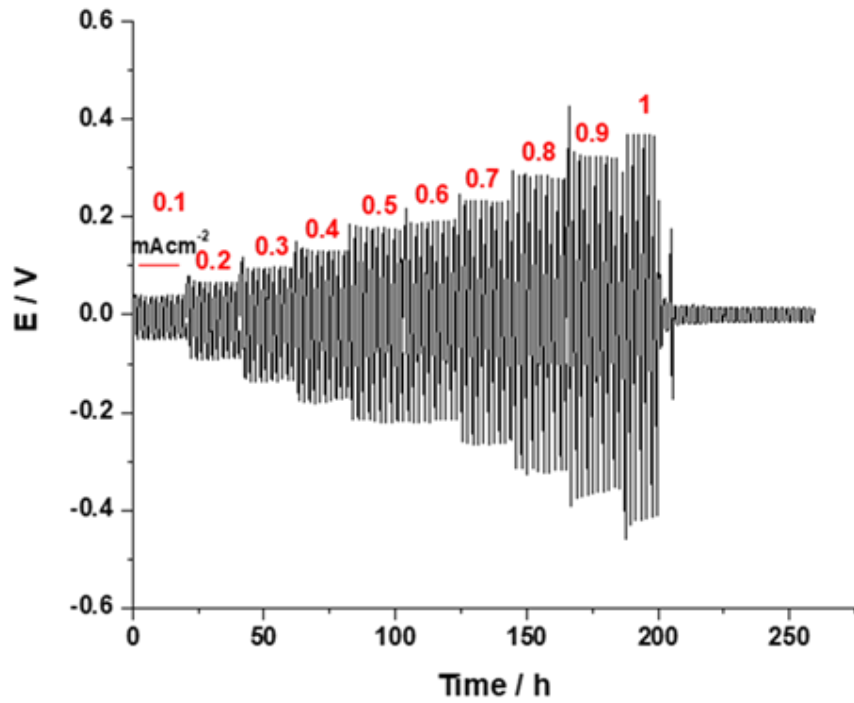
↳ Higher ionic conductivity & better performance

# TriBlock Copolymer Electrolytes

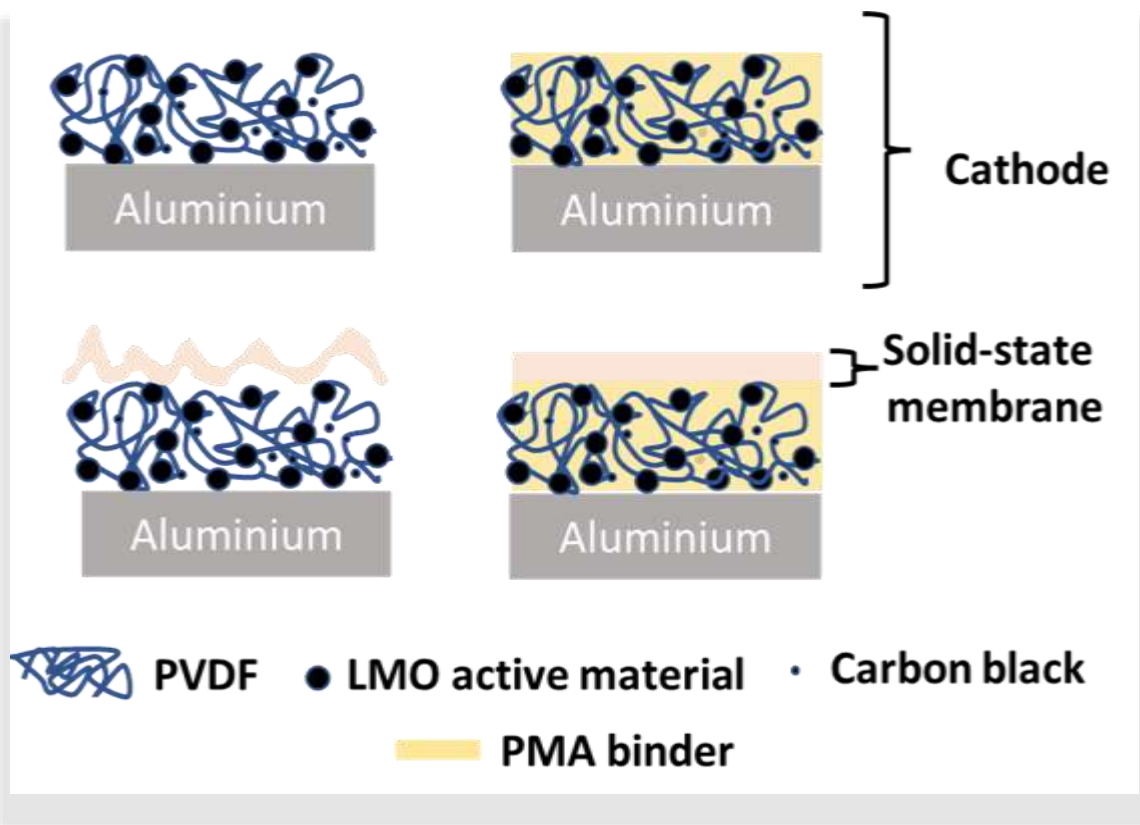
- Additional mechanical block To improve the mechanical properties of the SPE
- To maintain mechanical properties at higher LiFSI and IL concentrations to support higher current densities in Li metal cycling.



# TriBlock Copolymer Electrolytes-Electrochemical properties

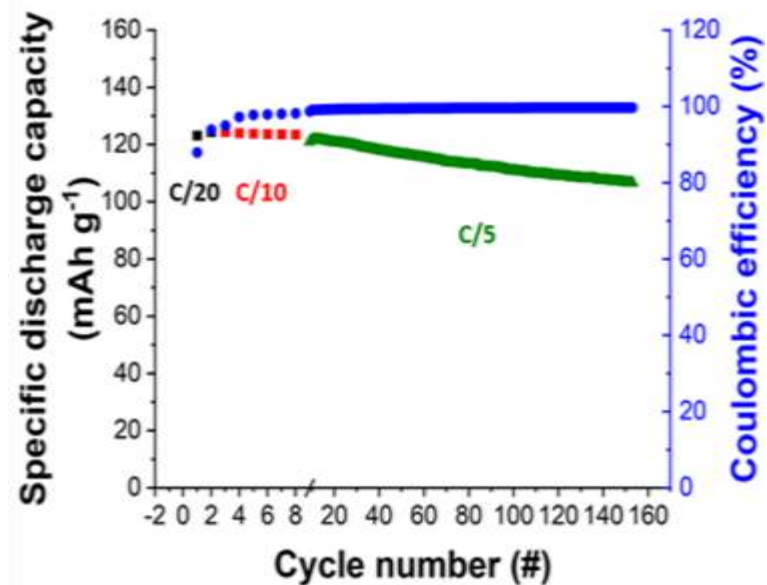


# All-solid-state Li metal battery

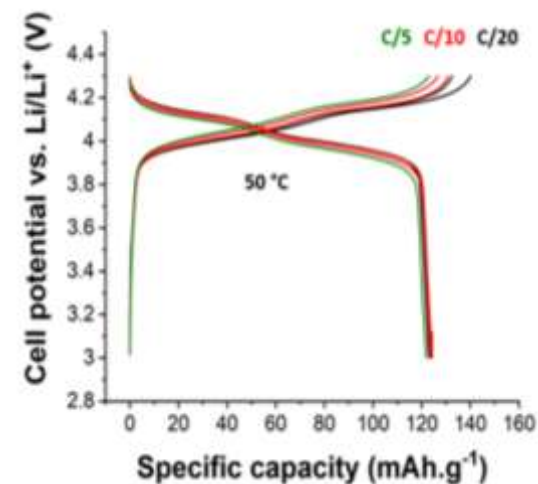
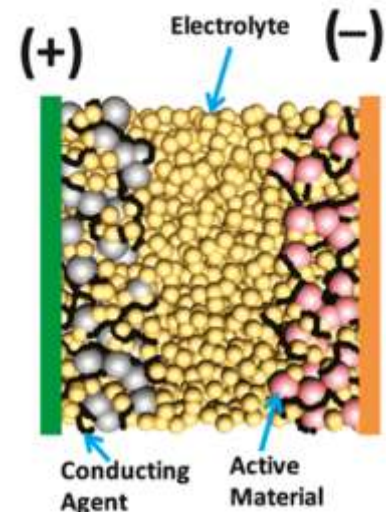


60wt% LMO / 30 wt% SPE / 5wt% PVDF / 5wt% C-black

Li | SPE | LiMn<sub>2</sub>O<sub>4</sub> (LMO)

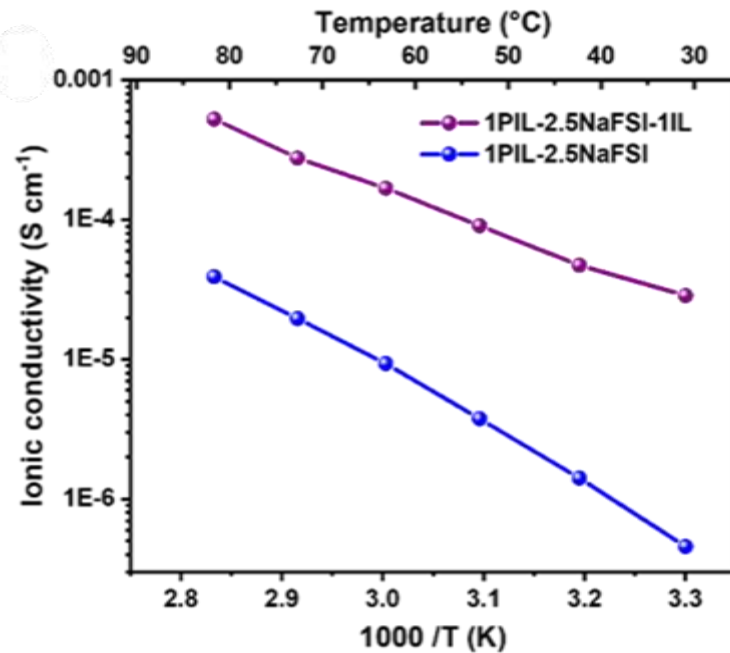
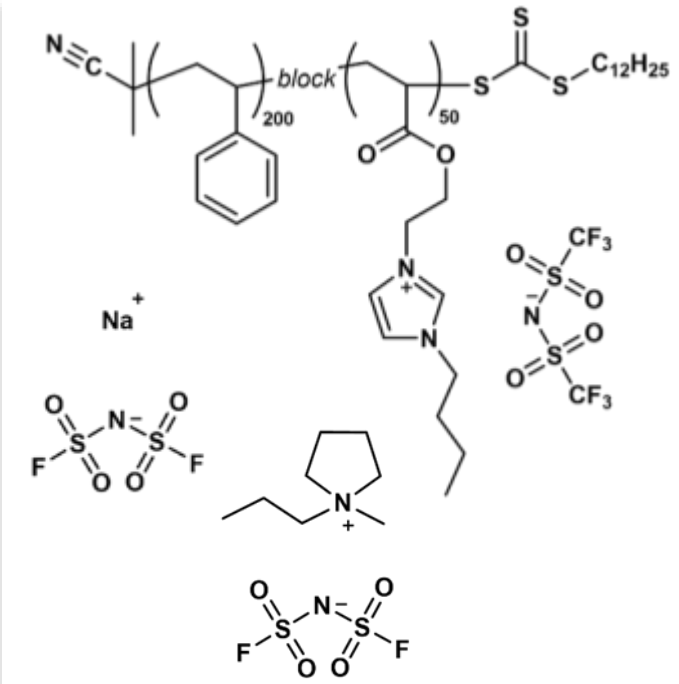


Bulk solid-state battery



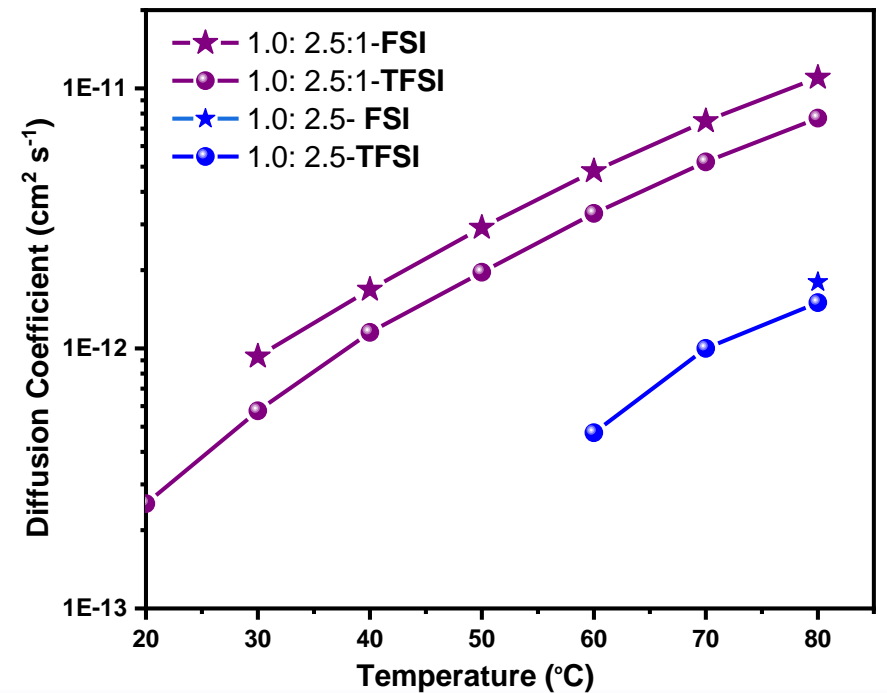
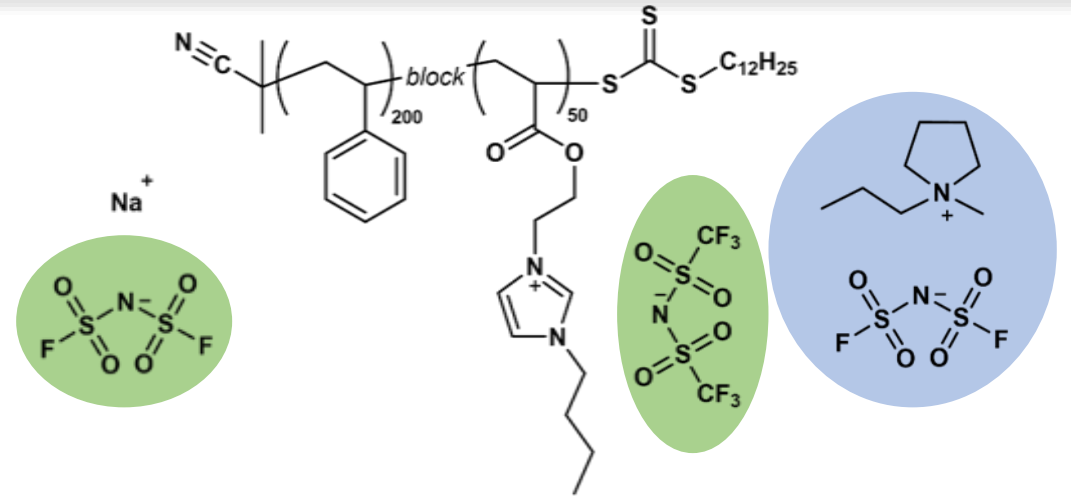
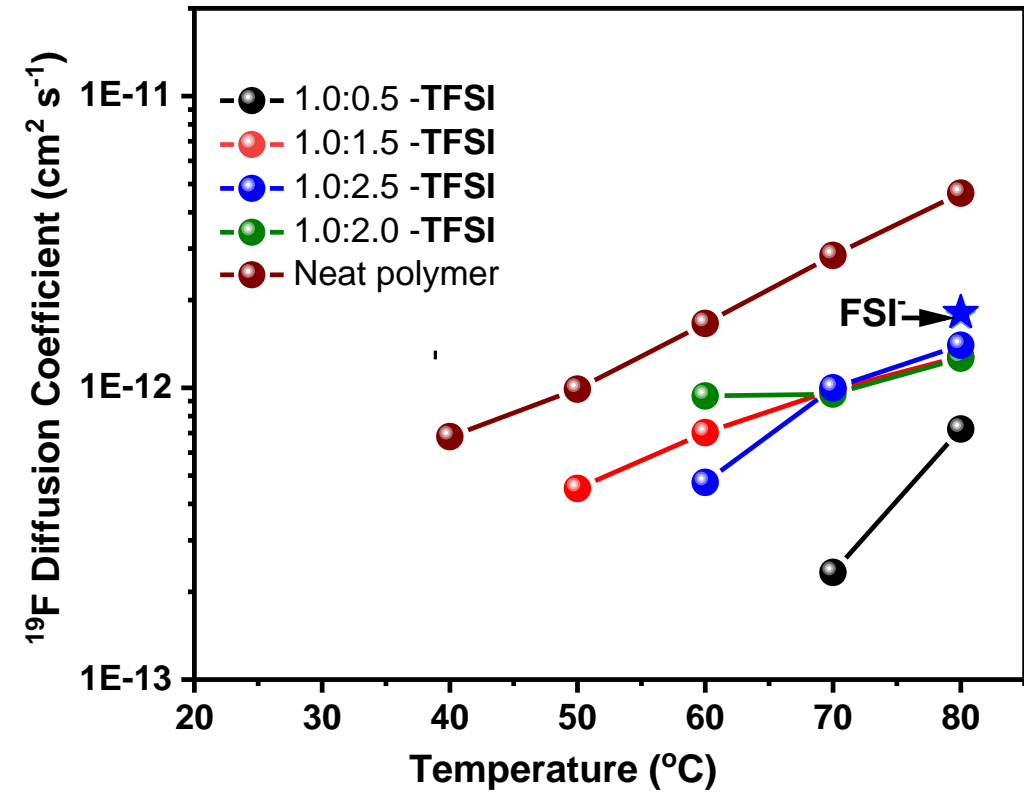


# Block Copolymer Electrolytes in *Na* devices



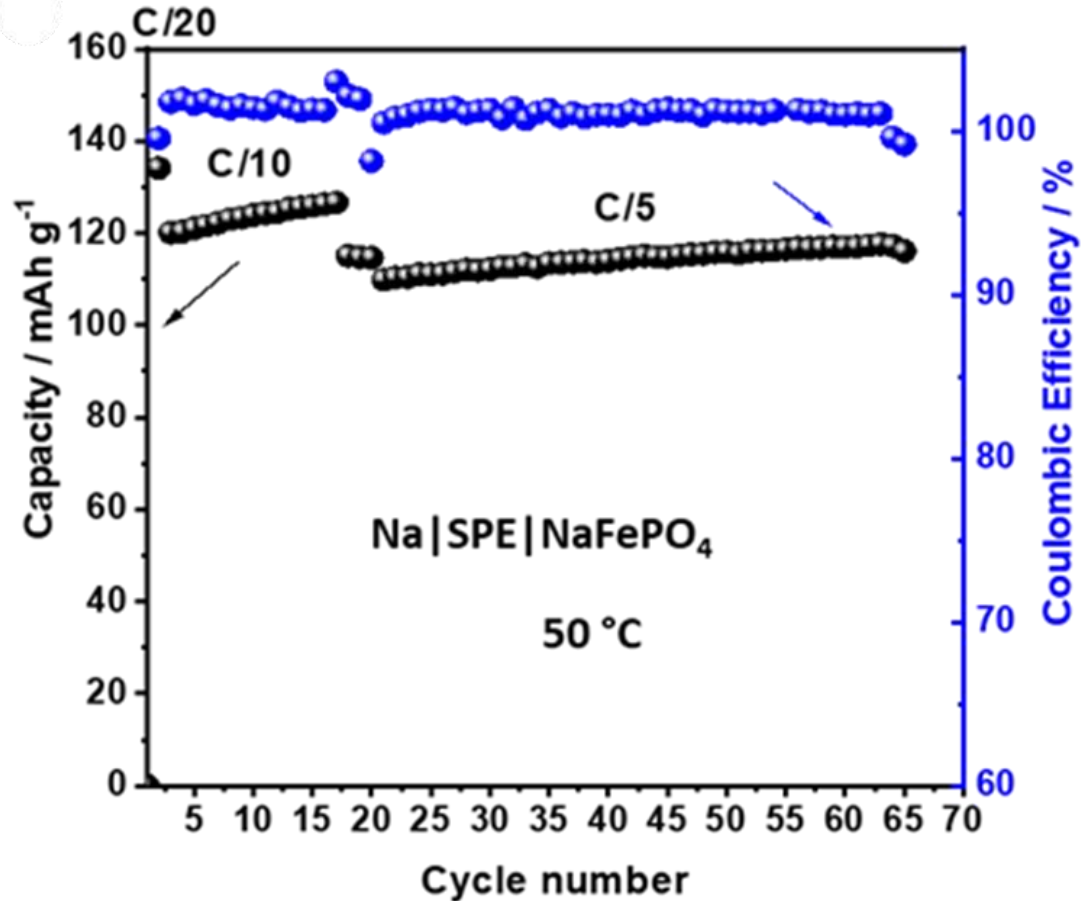
PIL:NaFSI: IL molar ratio	T <sub>g</sub> (°C) Onset	ΔT <sub>g</sub> (°C)	Conductivity at 70 °C (S cm <sup>-1</sup> )
1.0:0.5:0.0	-11	15	1.8 x 10 <sup>-6</sup>
1.0:1.5:0.0	-19	16	6.6 x 10 <sup>-6</sup>
1.0:2.0:0.0	-19	16	5.3 x 10 <sup>-6</sup>
1.0:2.5:0.0	-23	15	2 x 10 <sup>-5</sup>
1.0:3.0:0.0	-21	14	2.1 x 10 <sup>-5</sup>
<b>1.0:2.5:1.0</b>	<b>-43</b>	<b>28</b>	<b>2.8 E-04</b>

# Block Copolymer Electrolytes in *Na* devices; Ion diffusions and correlations



Dani Kourati poster #25

# All-solid-state Na metal battery

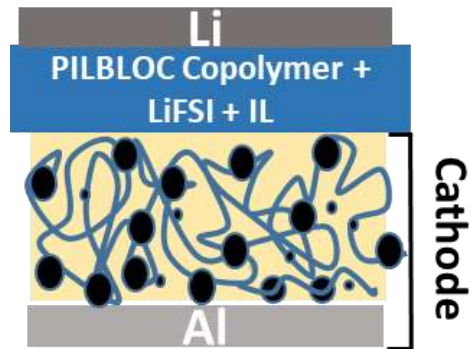


- An initial discharge capacity of 134 at C/20 and 120 and 115 mAh g<sup>-1</sup> at C/10 and C/5 at 50 °C.
- Demonstrating favourable charge–discharge performance, maintaining a capacity of 117 mAh g<sup>-1</sup> for 60 cycles at C/5 at 50 °C which is similar than the initial capacity,
- Exhibiting a high coulombic efficiency of 99.5% at the end of 60 cycles.

# Conclusion

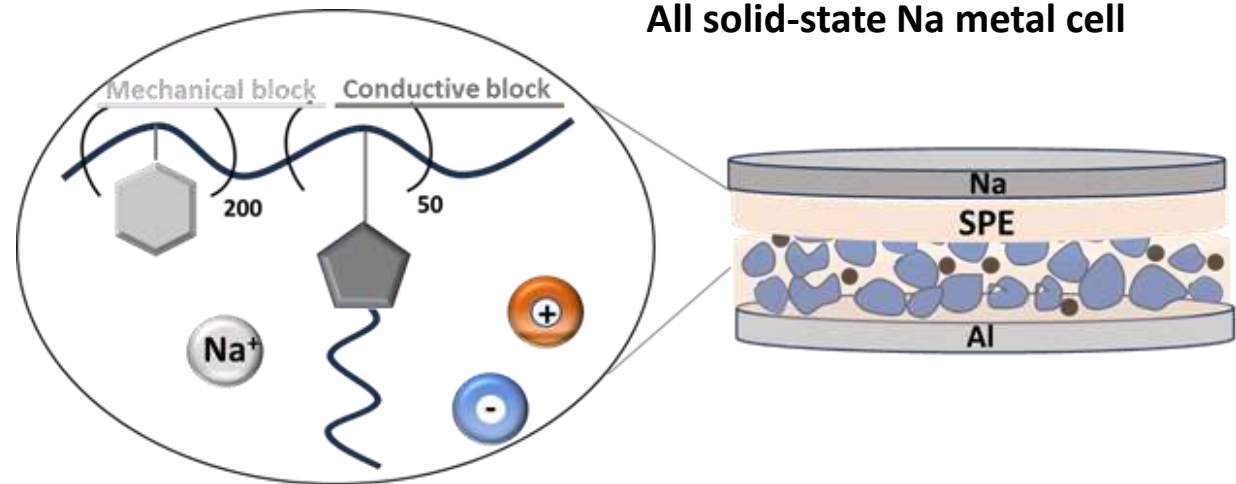
- Diblock and triblock polymer electrolytes with high mechanical stability and ionic conductivity
- Promising materials for high-voltage solid-state lithium and sodium-metal batteries

## All solid-state Li metal cell



-  PVDF
-  LMO active material
-  PMA binder
-  Carbon black

## All solid-state Na metal cell





# Acknowledgment

*Prof. Maria Forsyth*  
*A Prof. Luke O' Dell*  
*Dr. Greg Walker*  
*Luis Miguel*  
*Dani Kourati*



Battery Research  
and Innovation Hub

