

# Chelator Free Radiolabeling of PEGylated Graphene Nanosheet

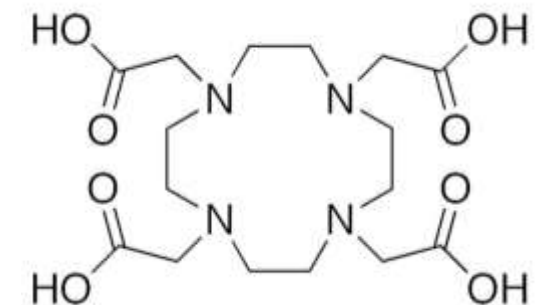
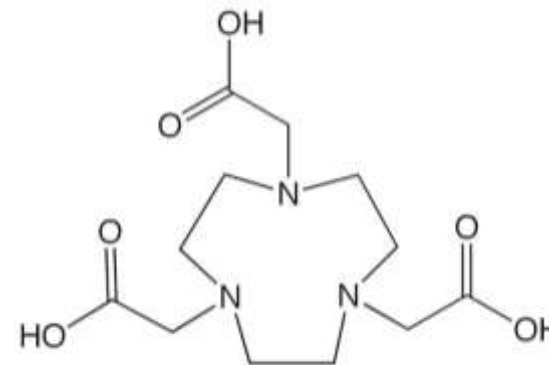
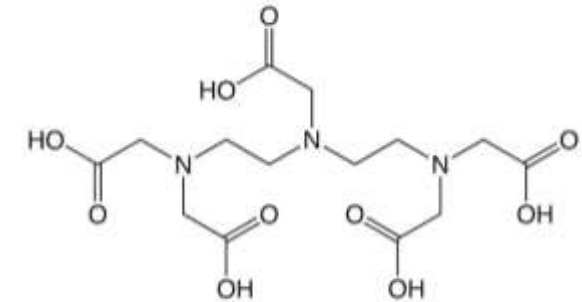
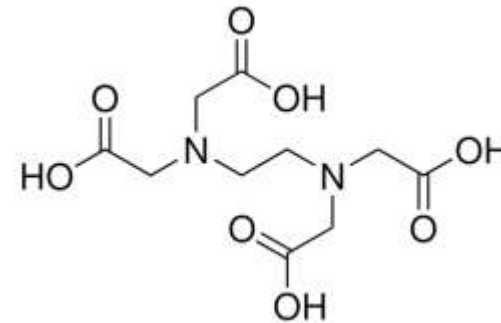
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# Chelators

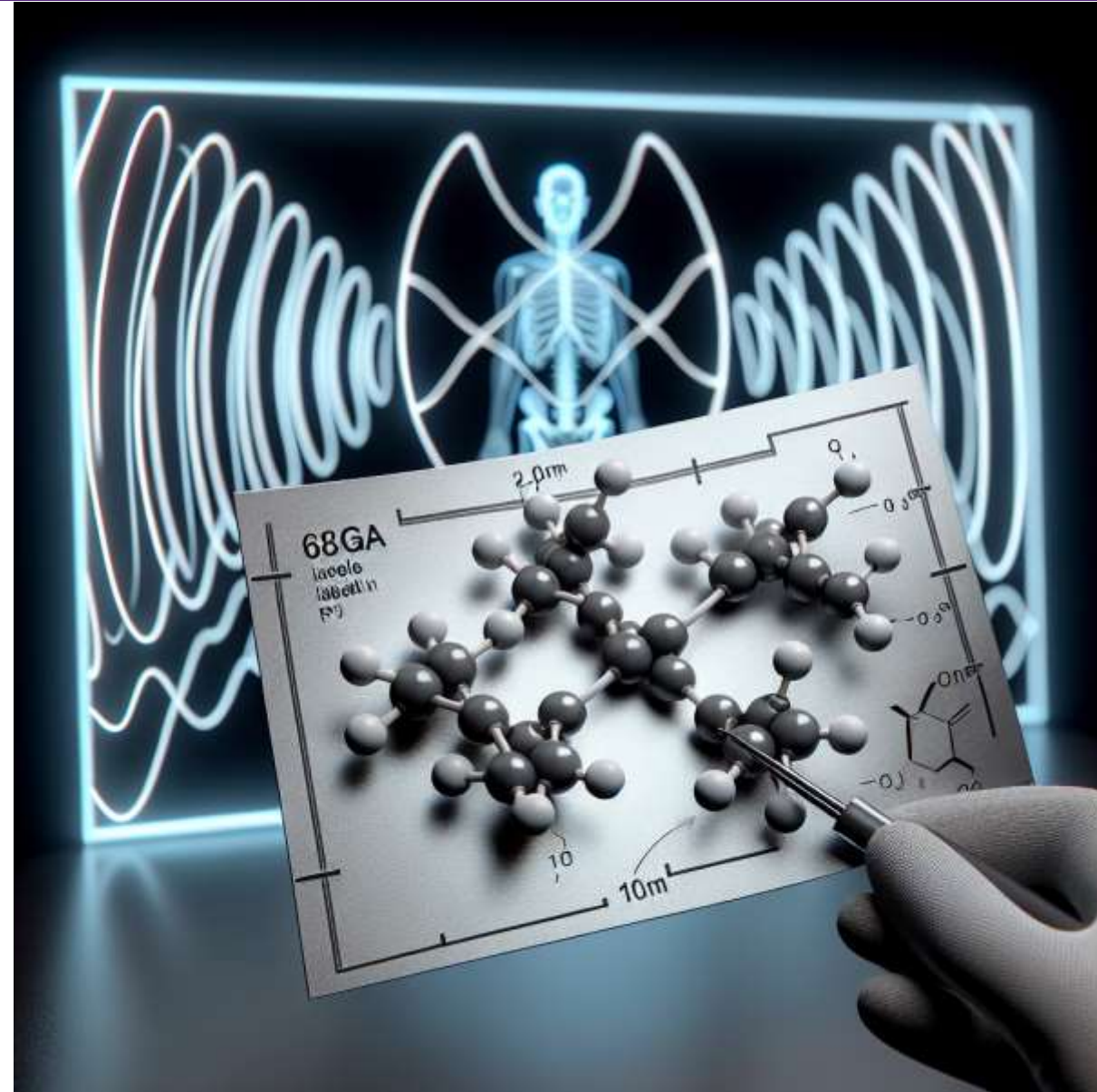
- Chelators are molecules that can form multiple bonds with a single metal ion, enhancing stability.
- They are used to capture and transport metal ions in biological and chemical systems.
- Common chelators include EDTA, DTPA, NOTA and DOTA, each with specific applications.
- Although chelators provide a stable linkage between the biomolecules/Nanoparticles and the radioactive isotope, they may also introduce certain limitations.
- Such as potential interference with the biological activity of the molecule, alteration of its pharmacokinetics, or increased immunogenicity.



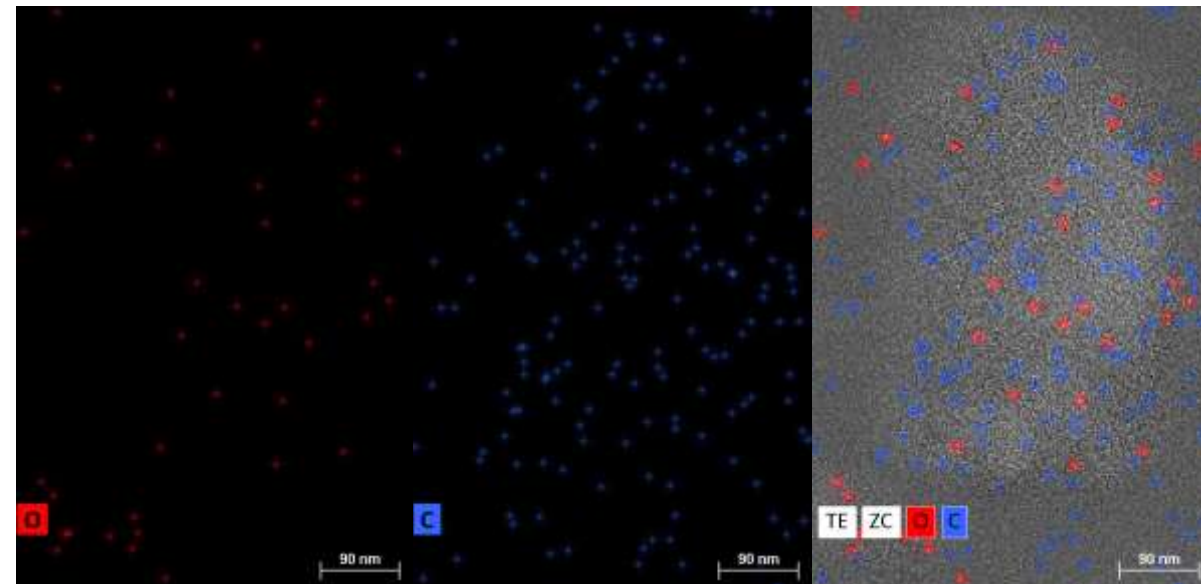
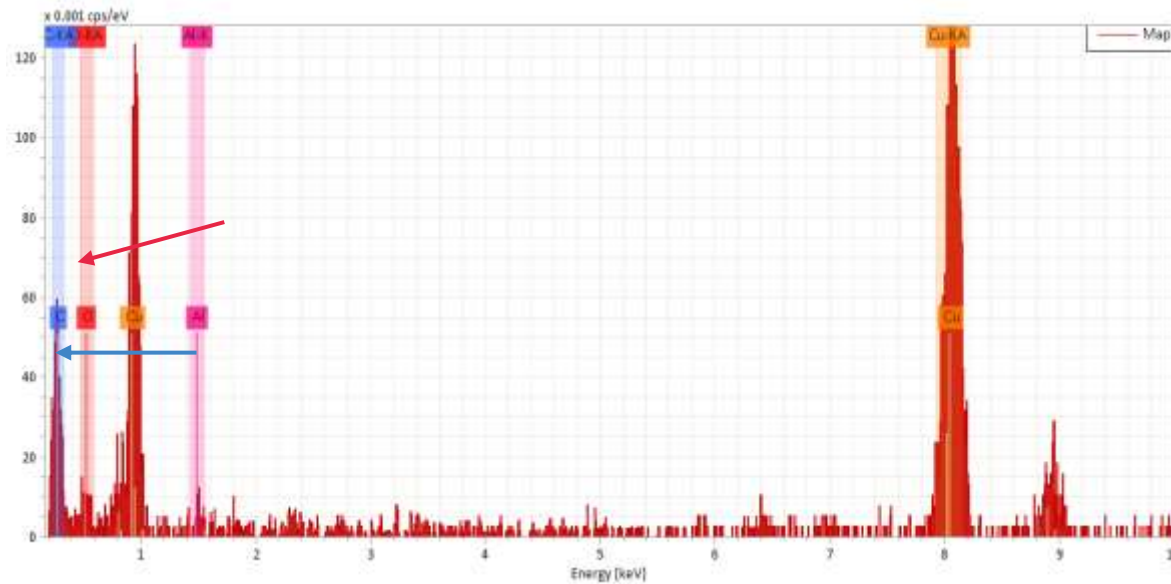
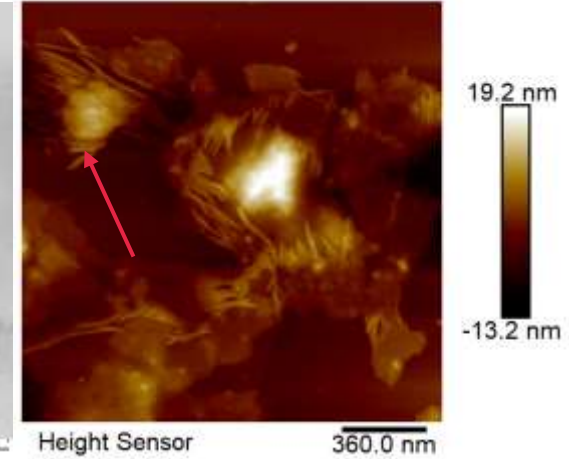
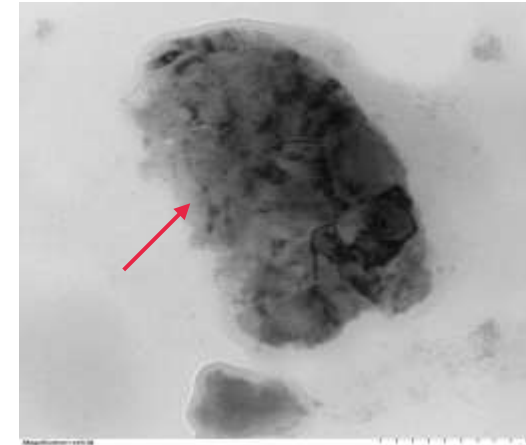
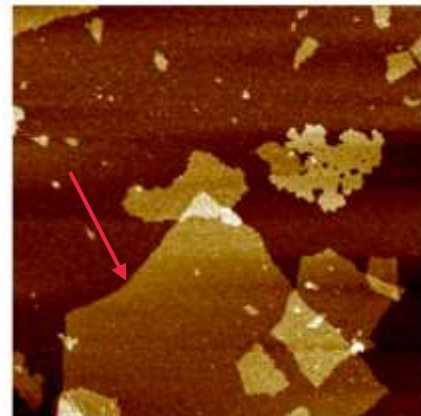
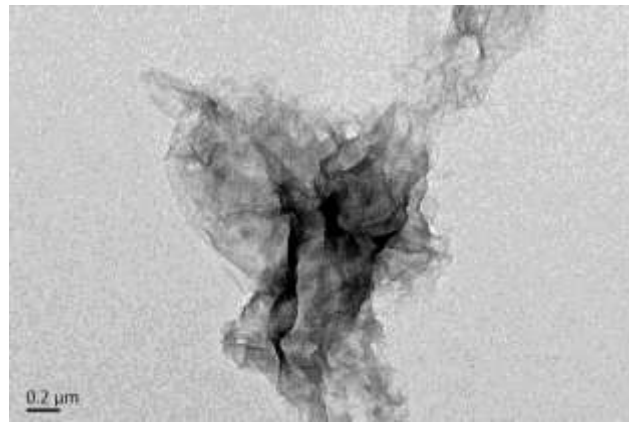
- Chelator-free radiolabelling refers to the process of directly labelling biomolecules/nanoparticles with radioisotopes without the use of a chelating agent.
- Strategies chelator-free radiolabelling, including:
  - **Direct labelling**
  - **Biorthogonal chemistry**
  - **Click chemistry**

## Why Chelator Free?

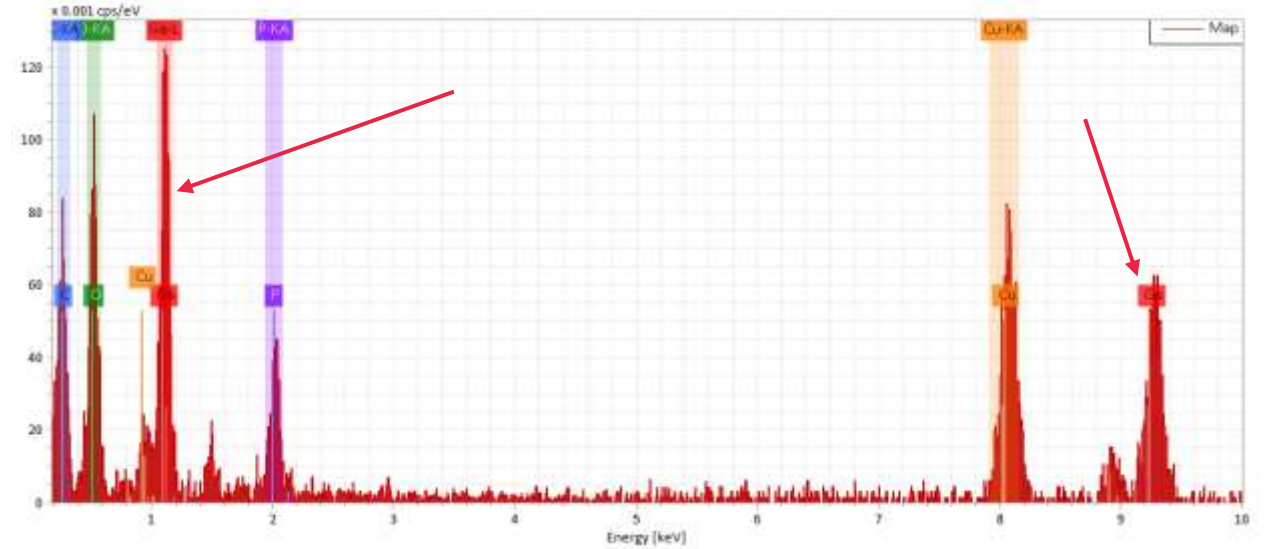
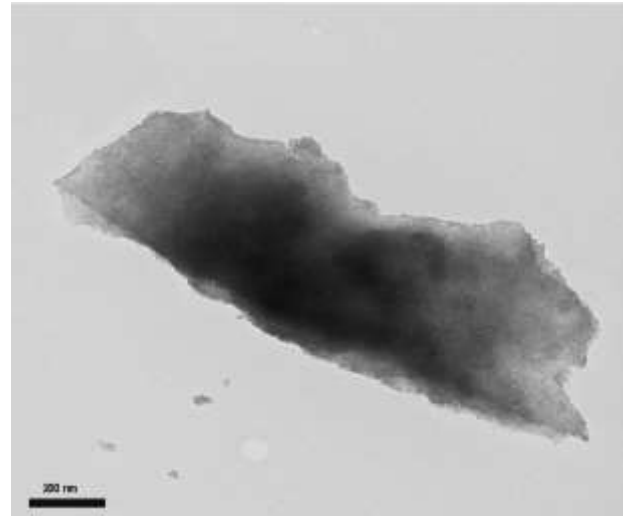
- **Advantages:** Simplified Chemistry, Improved Pharmacokinetics, Decreased Immunogenicity , Cost Effectiveness etc.



# Modification Graphene Oxide Nanosheets

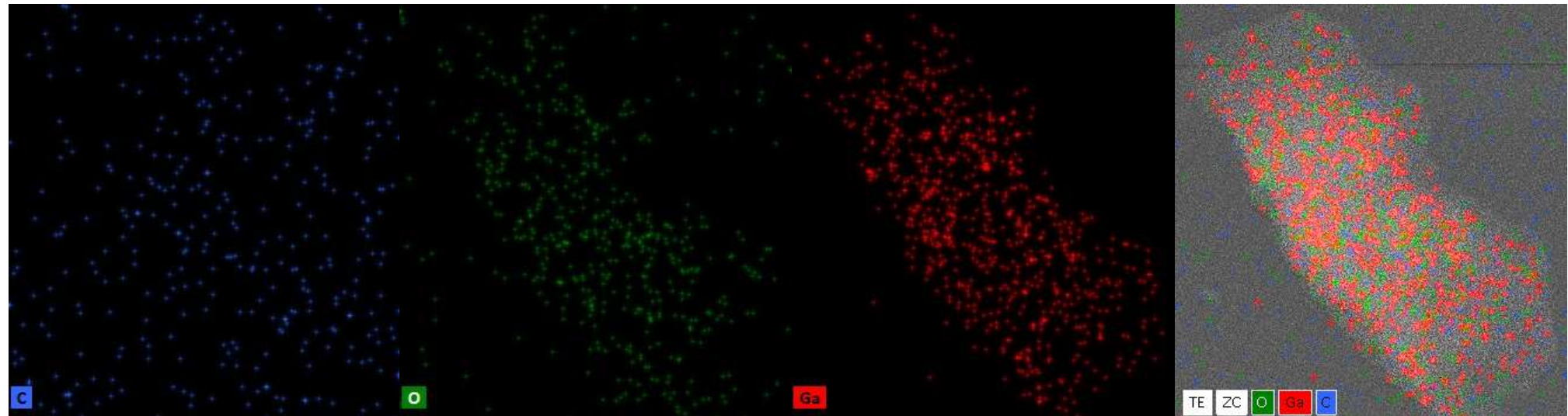


TEM Image of Graphene Oxide Nanosheet Doped with Gallium (Ga)



EDS of GO-PEG-Ga

TEM Elemental images of GO-PEG-Ga



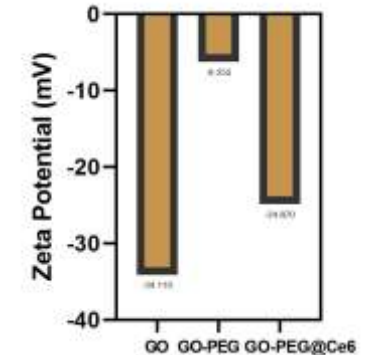
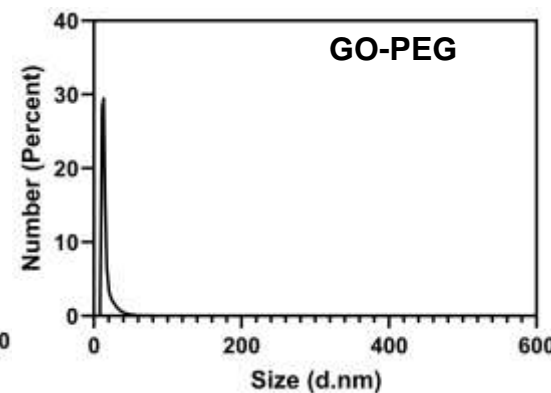
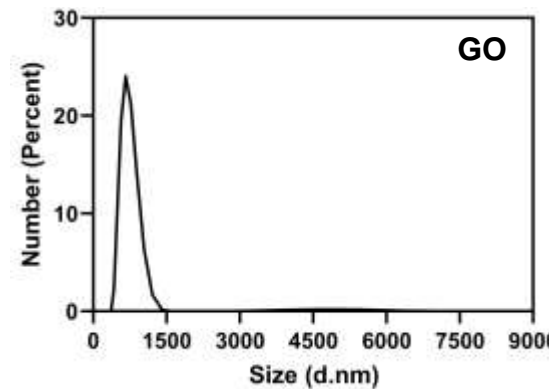
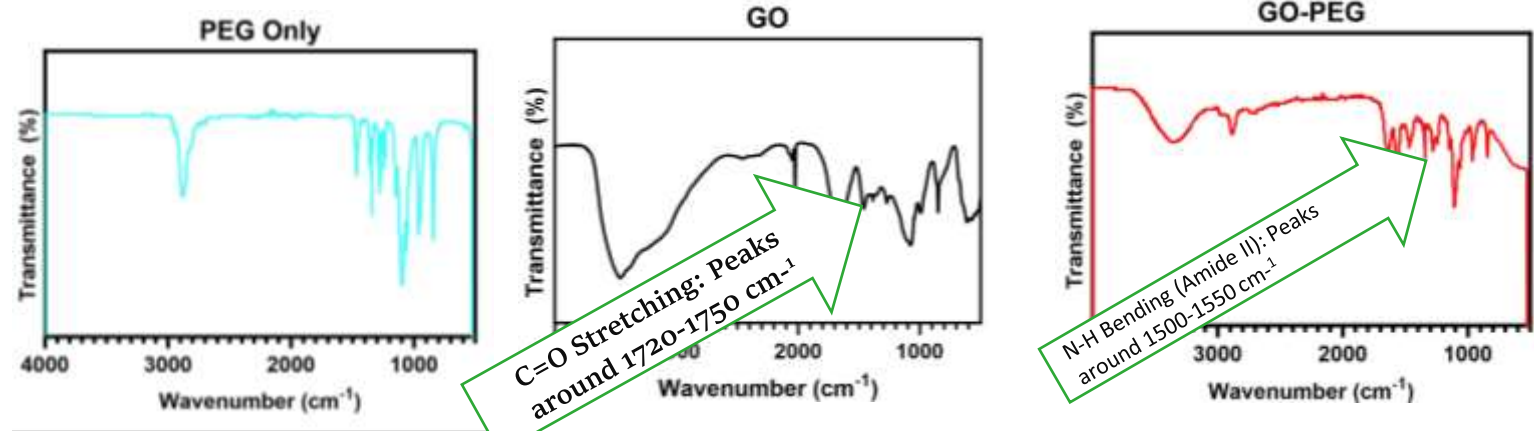
# Characterization of Graphene Oxide Nanosheets

**C=O Stretching:** Peaks around 1720-1750  $\text{cm}^{-1}$

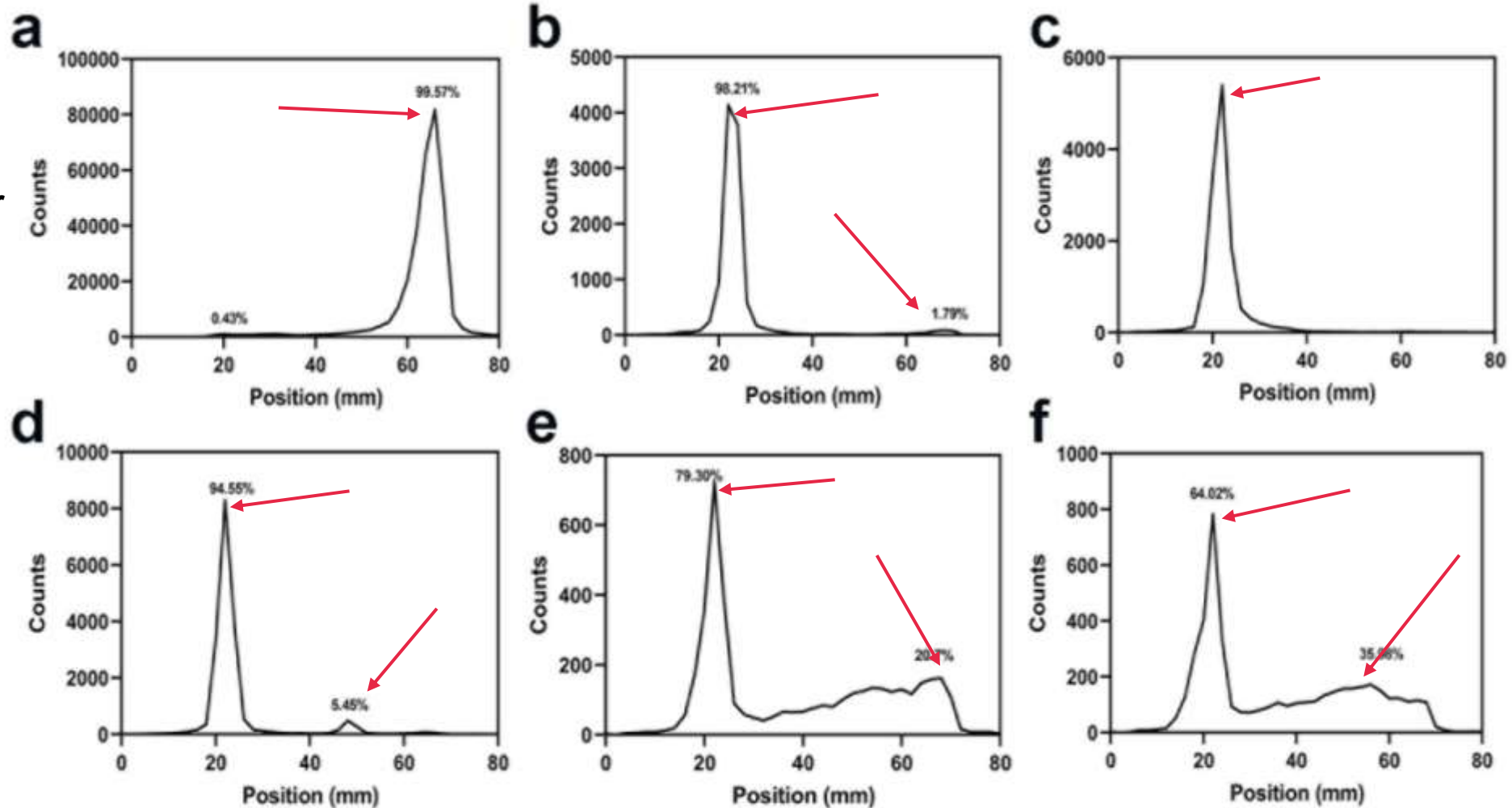
**N-H Bending (Amide II):** Peaks around 1500-1550  $\text{cm}^{-1}$

**PEG Backbone:** Additional peaks in the region of 1100-1300  $\text{cm}^{-1}$  corresponding to the stretching vibrations of C-O-C bonds in the PEG backbone.

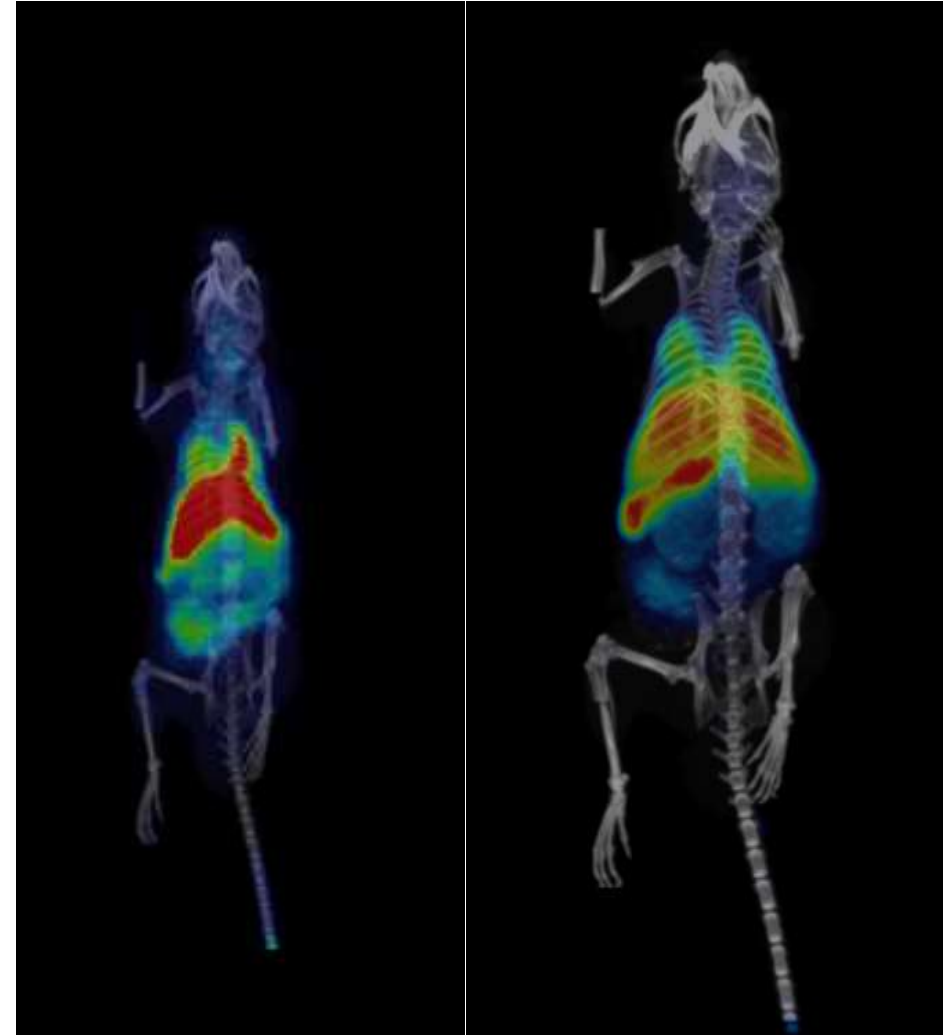
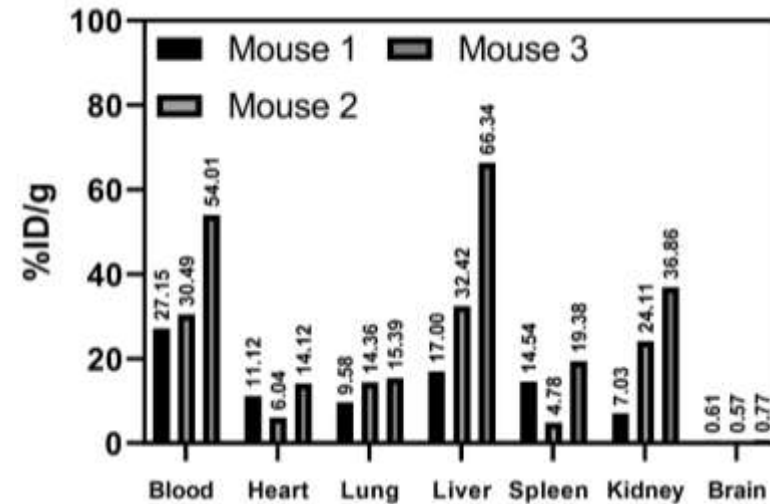
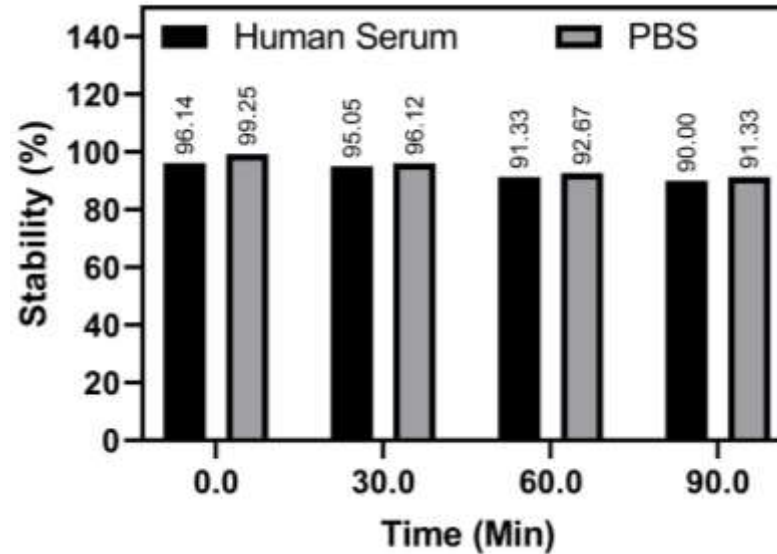
**Methoxy Group:** Peaks around 2850-3000  $\text{cm}^{-1}$  may indicate the stretching vibrations of C-H bonds in the methoxy group (-OCH<sub>3</sub>) of the methoxyamine PEG



- a) TLC of  $^{68}\text{Ga}$  in 50mM EDTA as control.
- b) TLC of GO-PEG-Ce6@ $^{68}\text{Ga}$  in Ammonium acetate buffer
- c) TLC of GO-PEG-Ce6@ $^{68}\text{Ga}$  in PBS.
- d) TLC of GO-PEG-Ce6@ $^{68}\text{Ga}$  in 50mM EDTA at 0 min.
- e) TLC of GO-PEG-Ce6@ $^{68}\text{Ga}$  in 50mM EDTA at 30 min.
- f) TLC of GO-PEG-Ce6@ $^{68}\text{Ga}$  in 50mM EDTA at 60 min

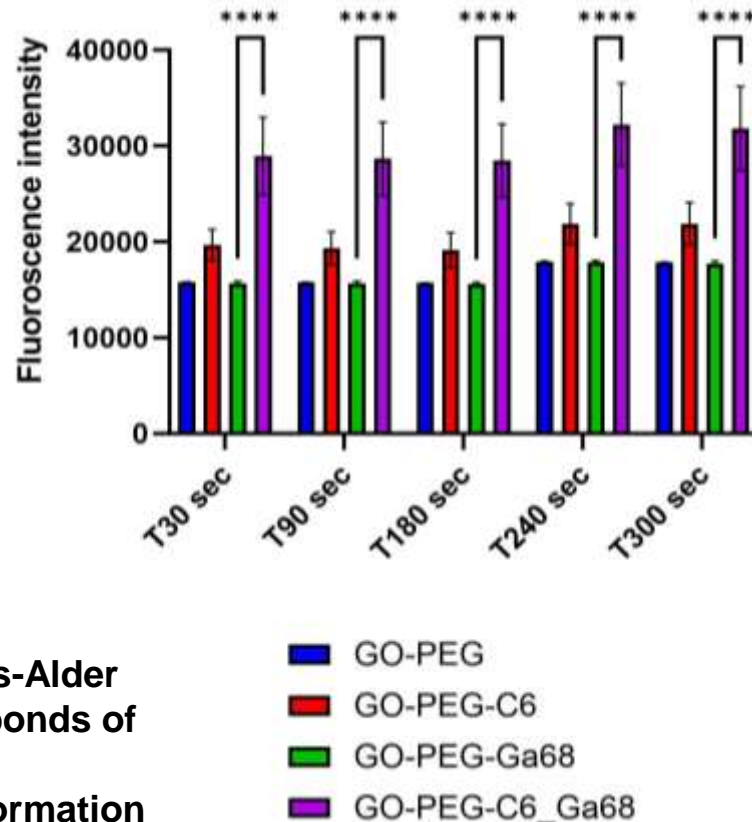
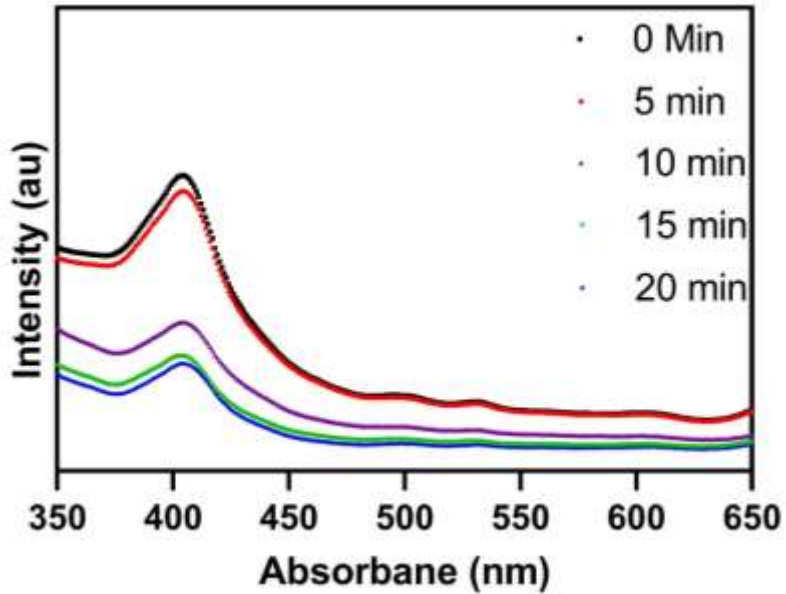


- ✓ The Radiolabelled Nanosheets displayed very good stability in Human Serum and PBS.
- ✓ In vivo PET Imaging showed material accumulation in liver, lungs, pancreases and in kidney.
- ✓ Ex Vivo Gamma counting





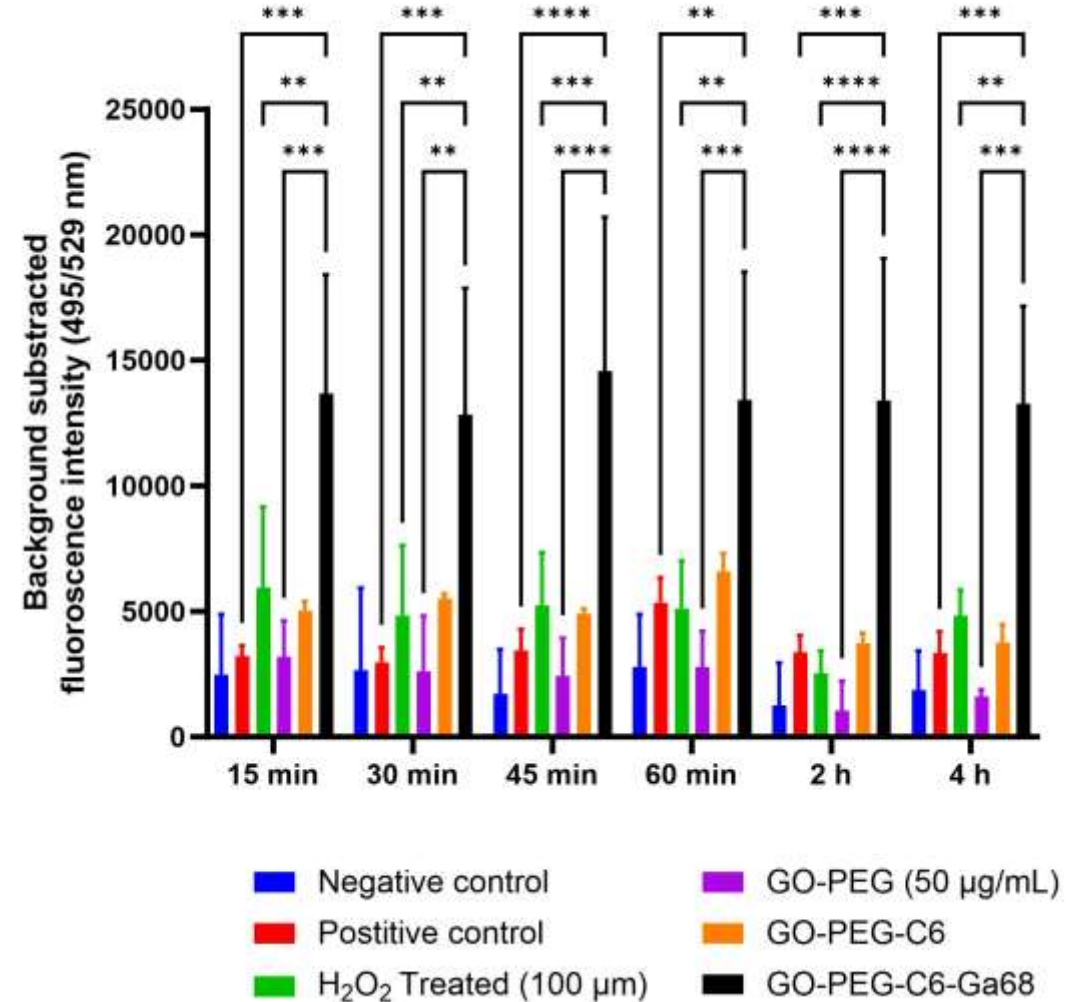
- ✓ Cherenkov radiation-induced energy transfer" likely refers to a process where the energy carried by Cherenkov radiation is transferred to another medium or particle.
- ✓ If a particles travel through a medium at a speed greater than the speed of light in that medium, Cherenkov radiation can be emitted.
- ✓ Radio isotopes induce CRIT when they undergo decay ( Alpha , Beta , Gamma )
  
- Bernhard, Y., Collin, B. & Decréau, R. Redshifted Cherenkov Radiation for in vivo Imaging: Coupling Cherenkov Radiation Energy Transfer to multiple Förster Resonance Energy Transfers. Sci Rep 7, 45063 (2017).  
<https://doi.org/10.1038/srep45063>



- ❑ The Singlet Oxygen Sensor Green reagent is highly selective for  $^1\text{O}_2$ .
- ❑ In the presence of singlet oxygen, it emits a green fluorescence like that of fluorescein
- ❑ (excitation/emission maxima ~504/525 nm).
- As shown in the graph, the  $^1\text{O}_2$  is generated only in the presence of both Ce6 and  $^{68}\text{Ga}$

- Singlet oxygen ( $^1\text{O}_2$ ) is generated, often through photosensitization reactions. ( $^1\text{O}_2$ ) undergoes a Diels-Alder cycloaddition reaction with the conjugated double bonds of DPBF.
- This reaction forms an endoperoxide product. The formation of the endoperoxide leads to a decrease in DPBF's absorbance, typically around 410 nm.
- Monitoring this decrease serves as a reliable method for detecting and quantifying singlet oxygen in chemical and biological systems.

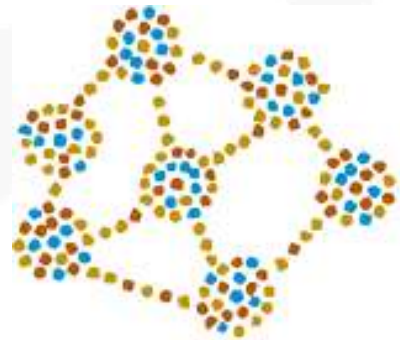
- **H<sub>2</sub>DCFDA, or 2',7'-dichlorodihydrofluorescein diacetate, is a non-fluorescence probe that can passively diffuse into cells.**
  - **Once inside the cell, the acetate groups of H<sub>2</sub>DCFDA are cleaved by intracellular esterases.**
  - **Finally, upon oxidation by ROS (hydroxyl, peroxy, and other species), the nonfluorescent H<sub>2</sub>DCFDA is converted to the highly fluorescent 2',7'-dichlorofluorescein (DCF).**
- ☐ **At any time point the ROS levels in cells treated with GO-PEG-C6-<sup>68</sup>Ga is at least 2.5-fold more than cells treated with ROS inducer (abcam) and 100 μmol H<sub>2</sub>O<sub>2</sub>.**



- ✓ **Developing more controls around the design of experiments .**
- ✓ **Cellular Toxicity Studies in different cell lines.**
- ✓ **Evaluating the efficiency of light independent CRIT induced PDT in *Ex vivo* and *in vivo* tumour models.**

# Acknowledgement

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Dr Arun Balaji  
Thurecht Group Members



National  
Imaging  
Facility



Centre for Advanced Imaging  
[cai.centre.uq.edu.au/study](http://cai.centre.uq.edu.au/study)

