



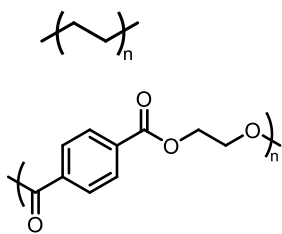
Biobased & Biocatalysed Covalent Adaptable Networks

Camille Bakkali-Hassani

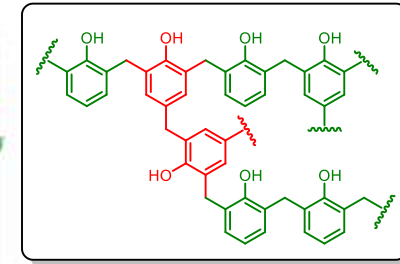
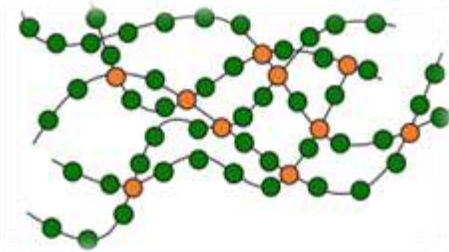
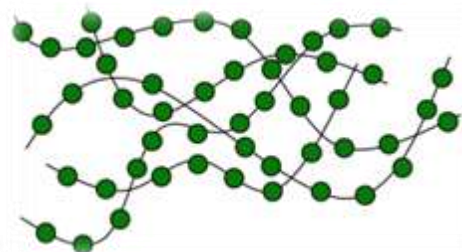


Covalent Adaptable Networks: A Third Family of Polymer

Thermoplastics



Linear polymer chains
 Flow upon heating ($T > T_g$ or T_m)
 Dissolve in good solvent



Crosslinked networks
 Do not flow nor dissolve
 (swell in good solvent)

Thermosets

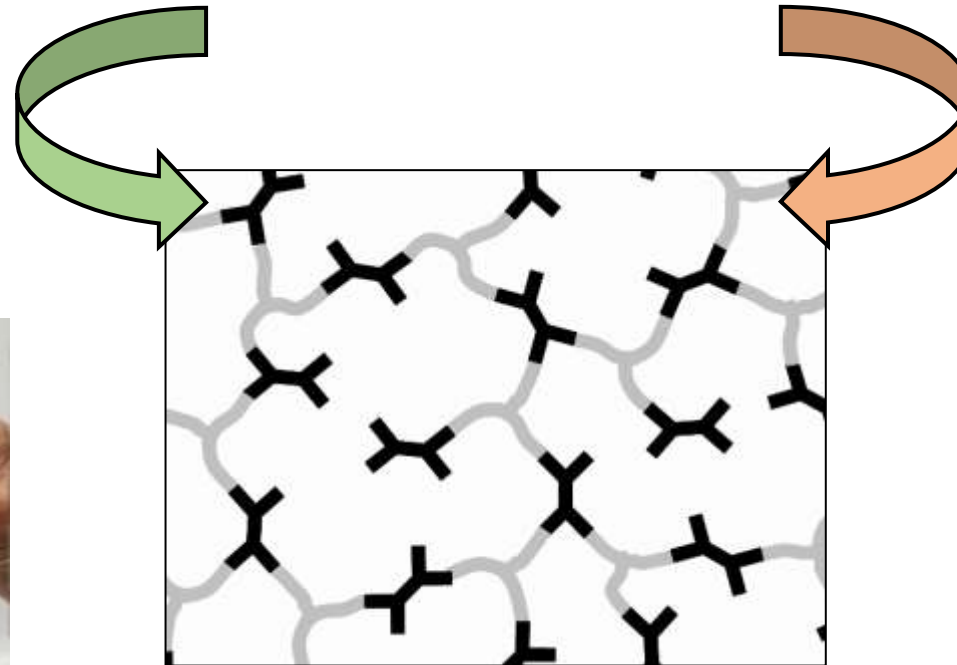
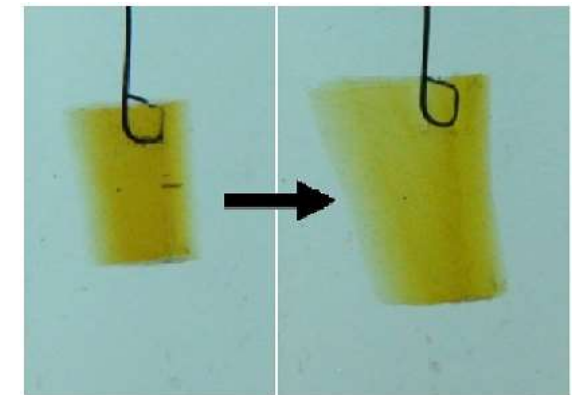
Bakelite®



Recyclable & Reshapable
Poor chemical resistance & mechanical properties

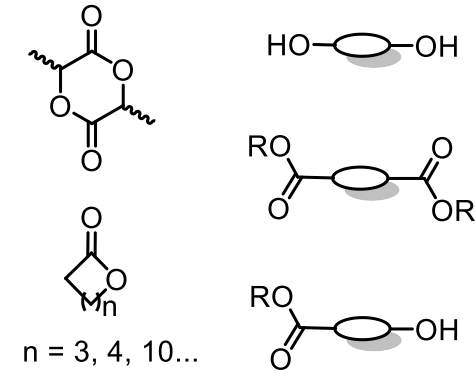
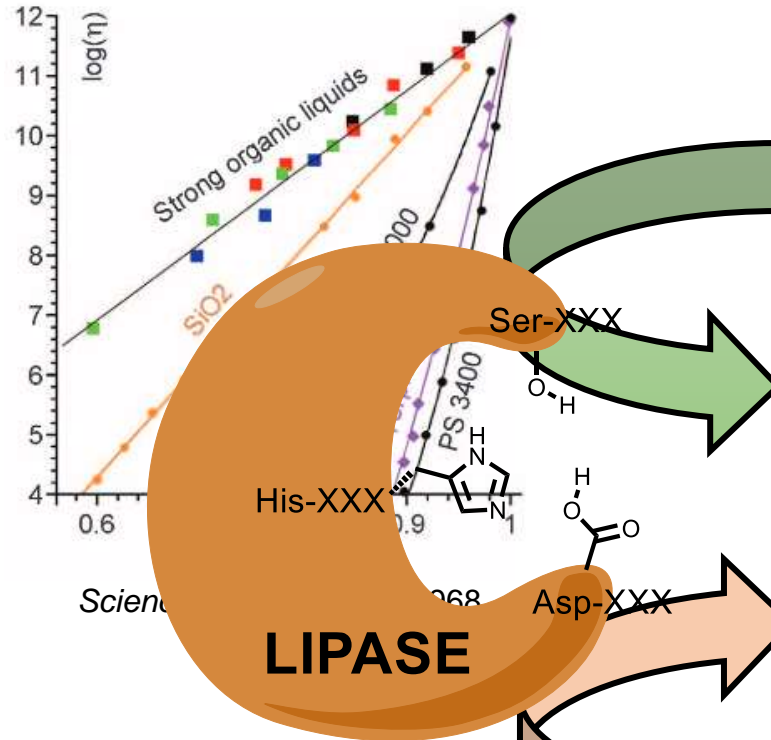
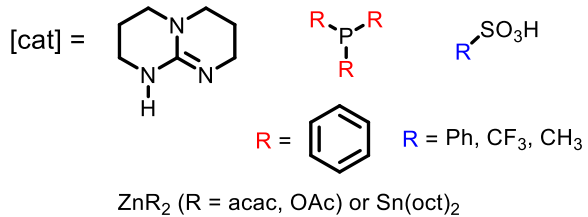
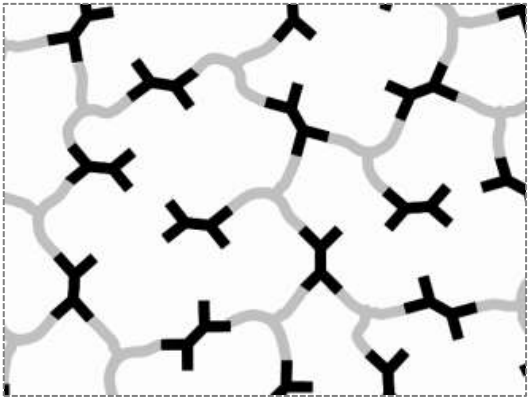
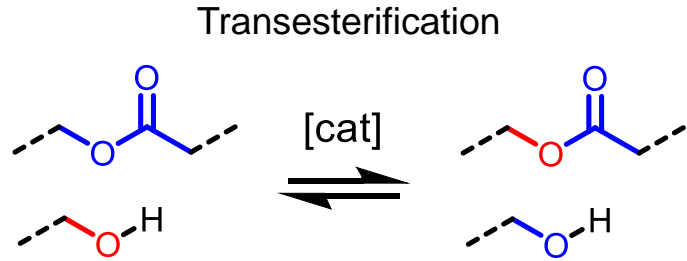


Not Recyclable & Reshapable
Excellent chemical resistance & mechanical properties



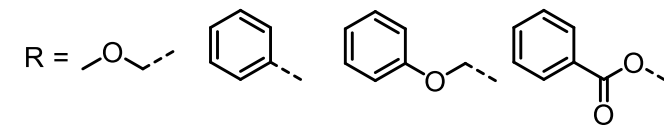
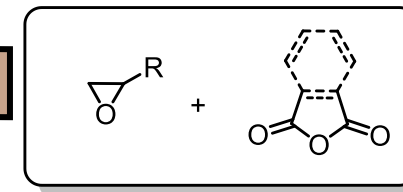
Covalent Adaptable Networks (CANs)

Transesterification-based vitrimers

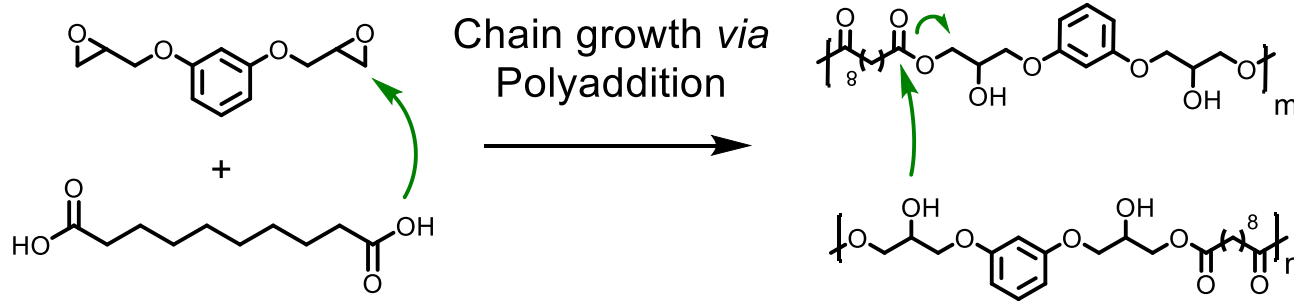


Ring-opening polymerisation
Step-growth polycondensation

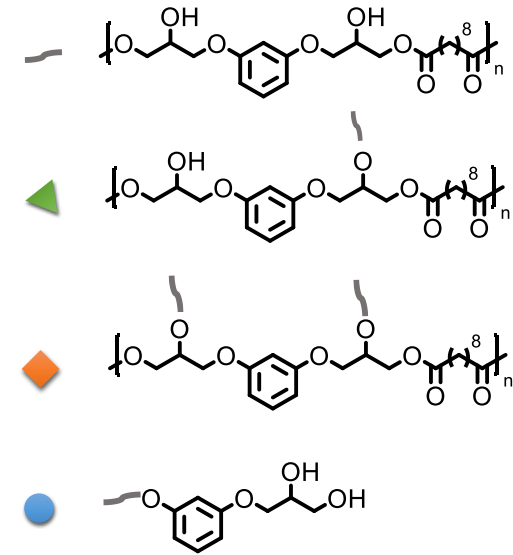
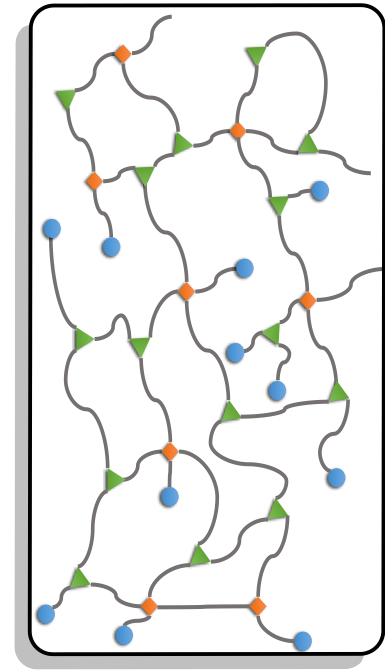
Epoxy-acyl addition
(is it truly enzymatic?)



Epoxy-Acid networks: polymerisation/crosslinking mechanism and features



Gelation via Transesterification



Gelation

$$\bar{f}_B = \frac{\sum_{i=0}^4 n_i \cdot i^2}{\sum_{i=0}^4 n_i \cdot i} = 2.5$$

\bar{f}_B : average functionality
 n_i : isomer probability
 i : isomer functionality

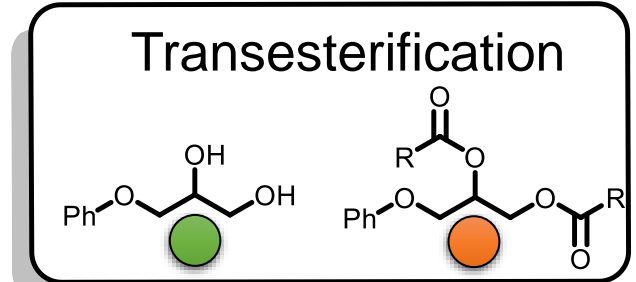
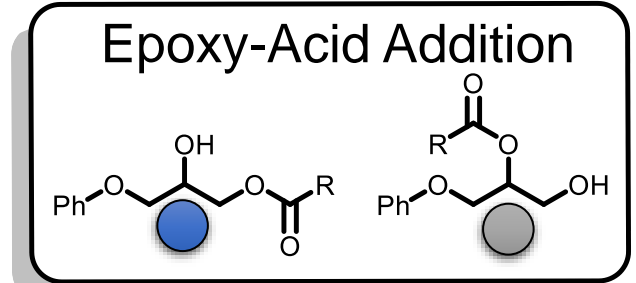
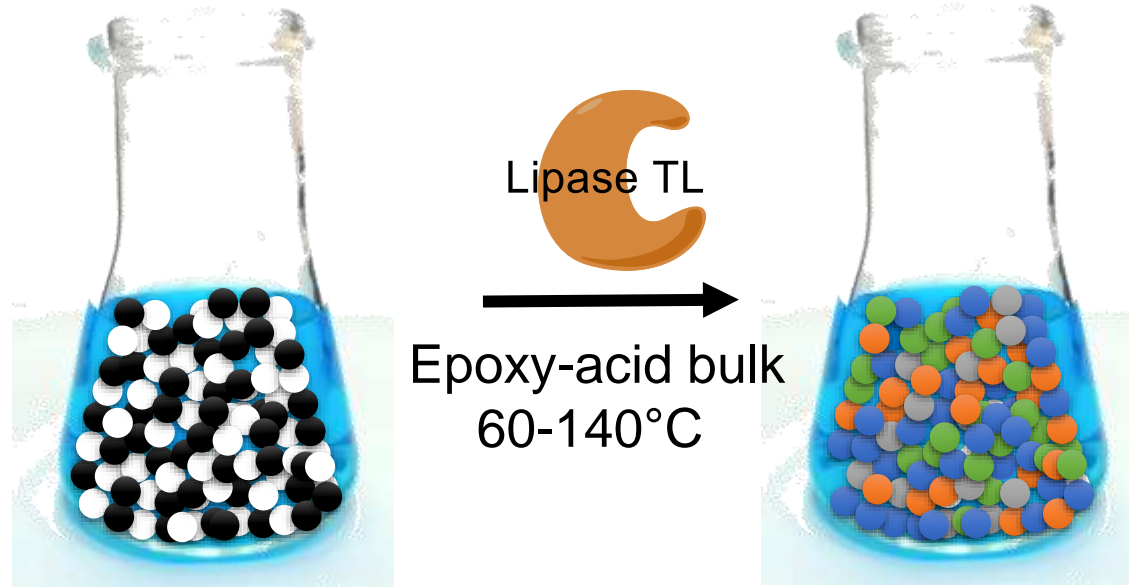
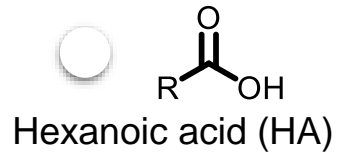
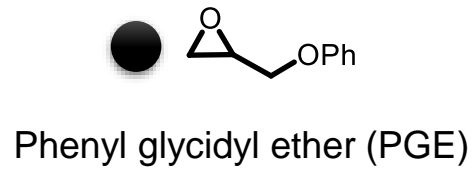
Classical catalyst

organic (imidazole-based, TBD, phosphines)
 organometallic (Sn-based, Zn-based etc.)

Conditions

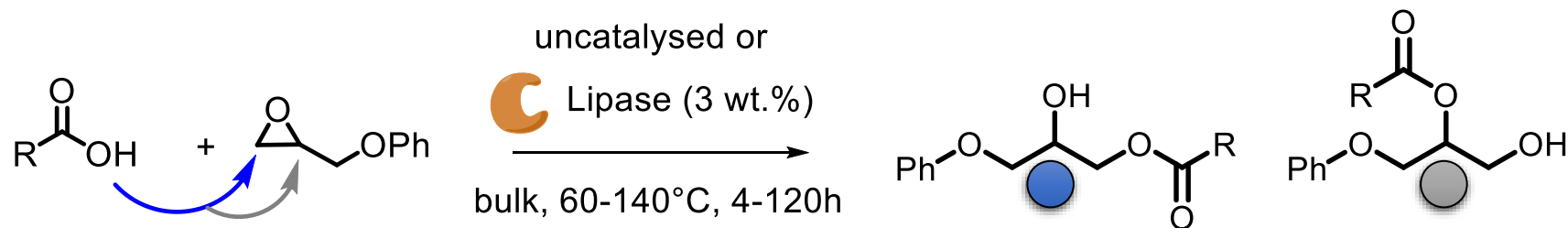
bulk (mixture of acid and epoxide)
 high temperature (**130°C-180°C**)

From model molecule studies to network build-up



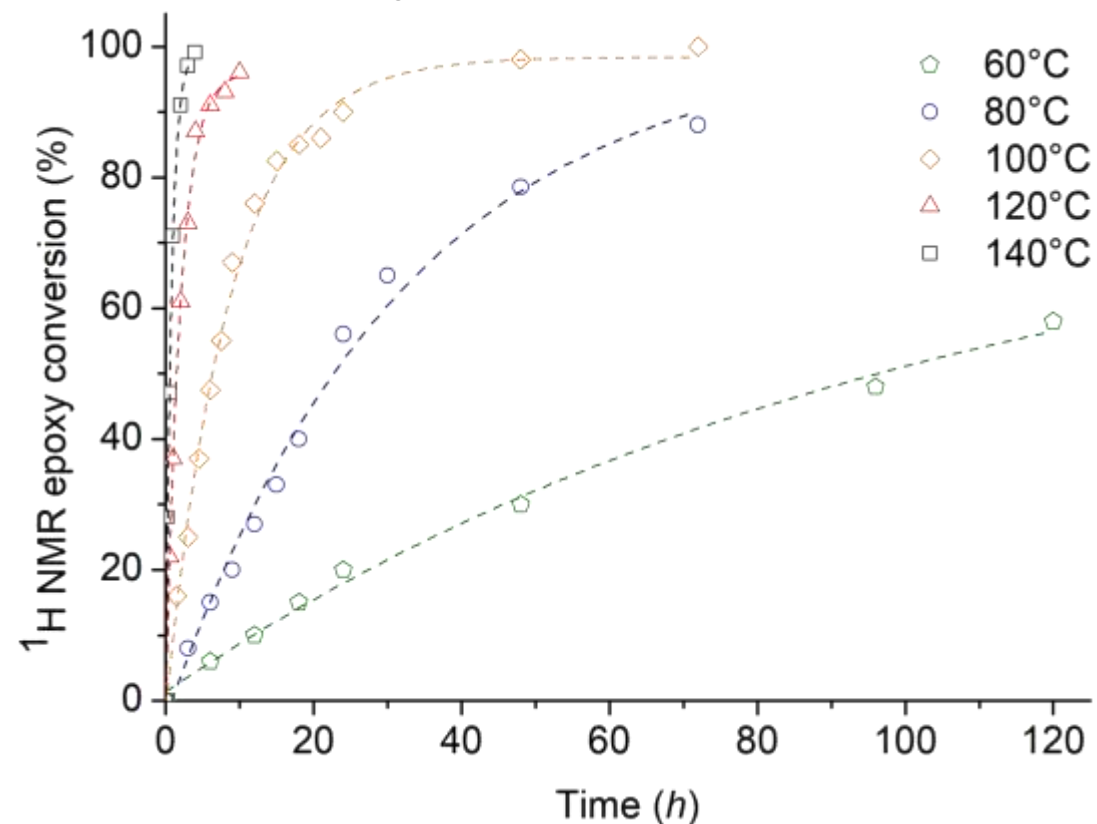
- 1- Find suitable conditions for material synthesis (T°C & water content)
- 2- Determine the mechanism (epoxy-acid addition and transesterification)
- 3- Is the enzyme still « alive » ?
- 4- Can we produce vitrimer or vitrimer-like materials ?

Model molecular reactions: Effect of temperature on epoxy-addition

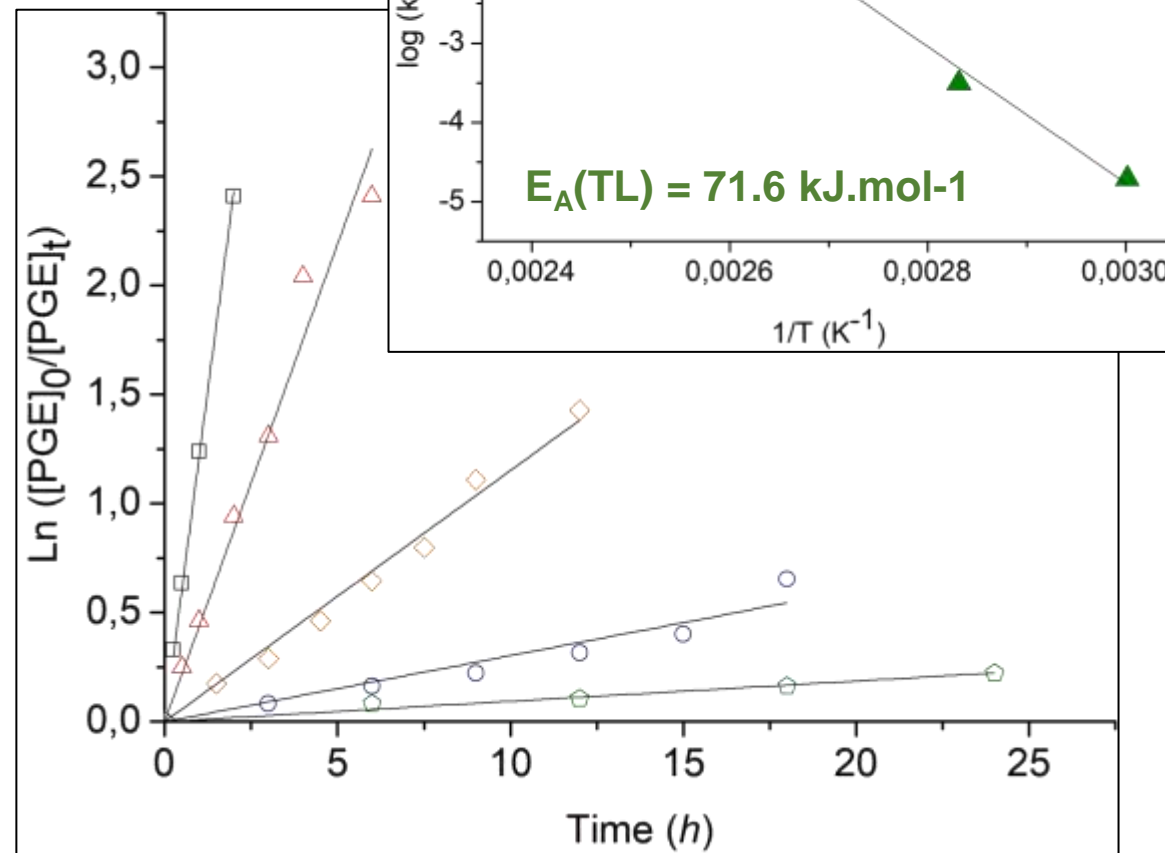
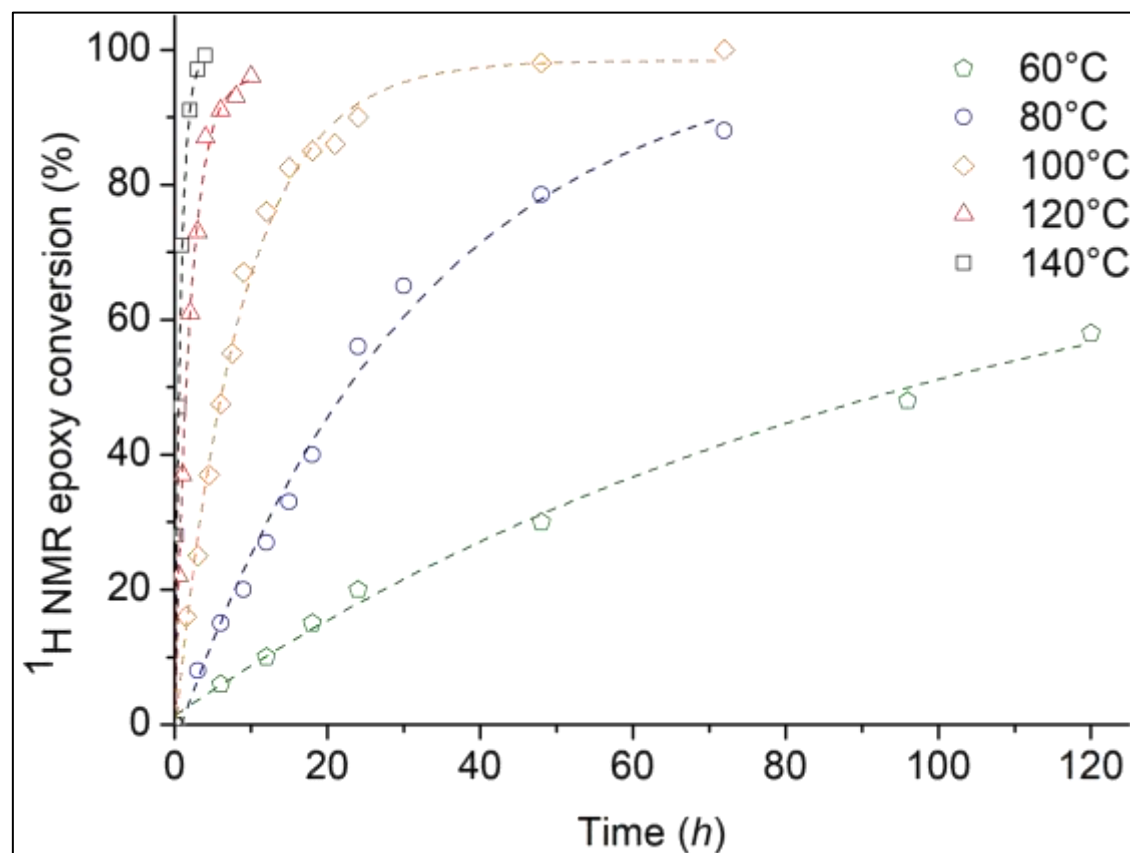
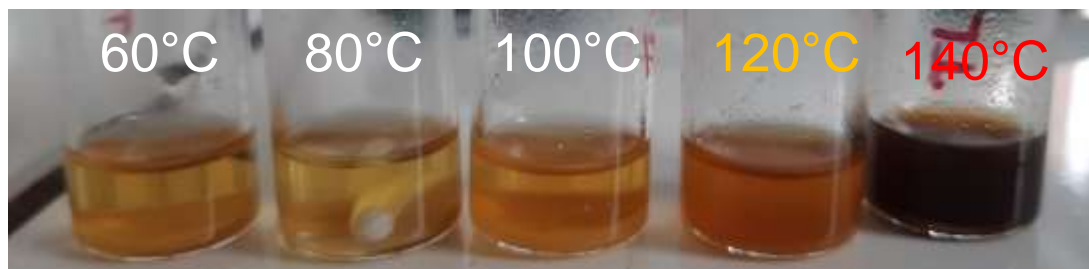


Label	Catalyst	T (°C)	Reaction time (in h)	¹ H NMR conv. (%)
NC60	none	60	120	2
TL60	Lipase TL	60	120	50
NC80	none	80	72	5
TL80	Lipase TL	80	72	88
NC100	none	100	48 (72)	30 (50)
TL100	Lipase TL	100	48	98
NC120	none	120	10	32
TL120	Lipase TL	120	10	≥ 99
NC140	none	140	4	37
TL140	Lipase TL	140	4	≥ 99

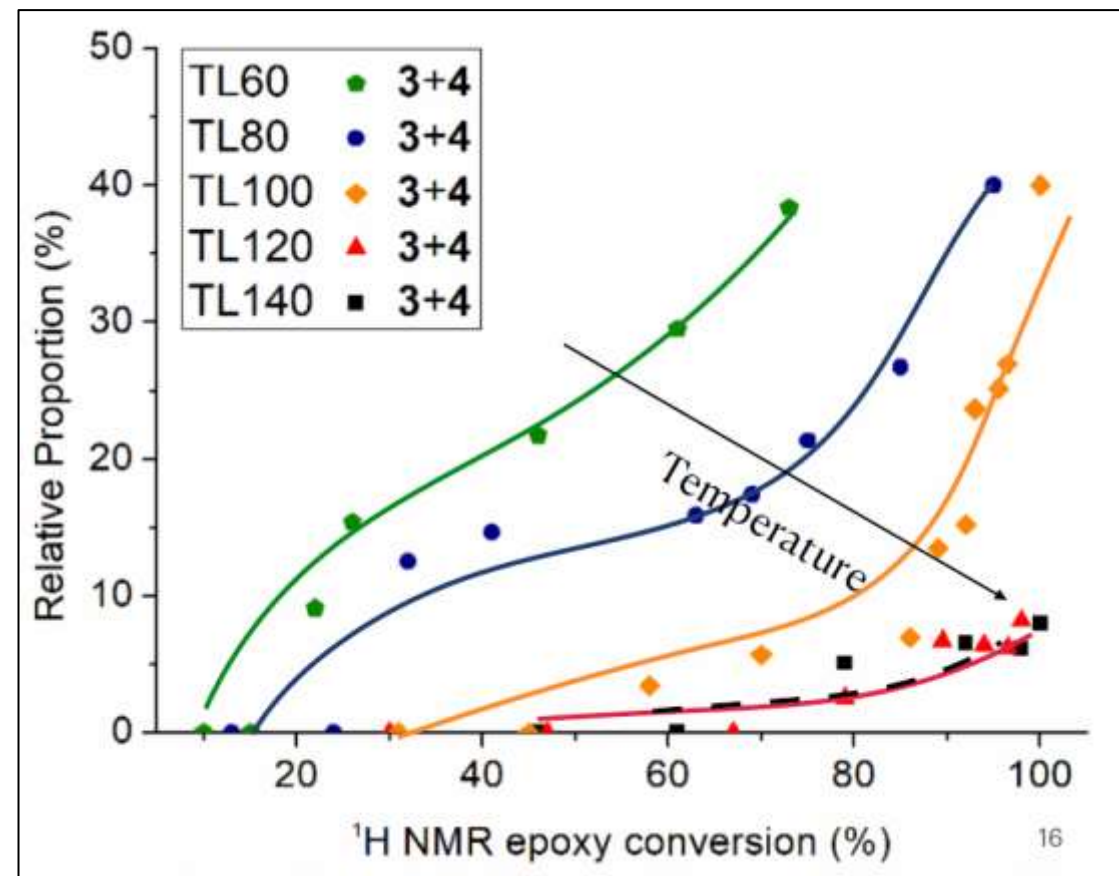
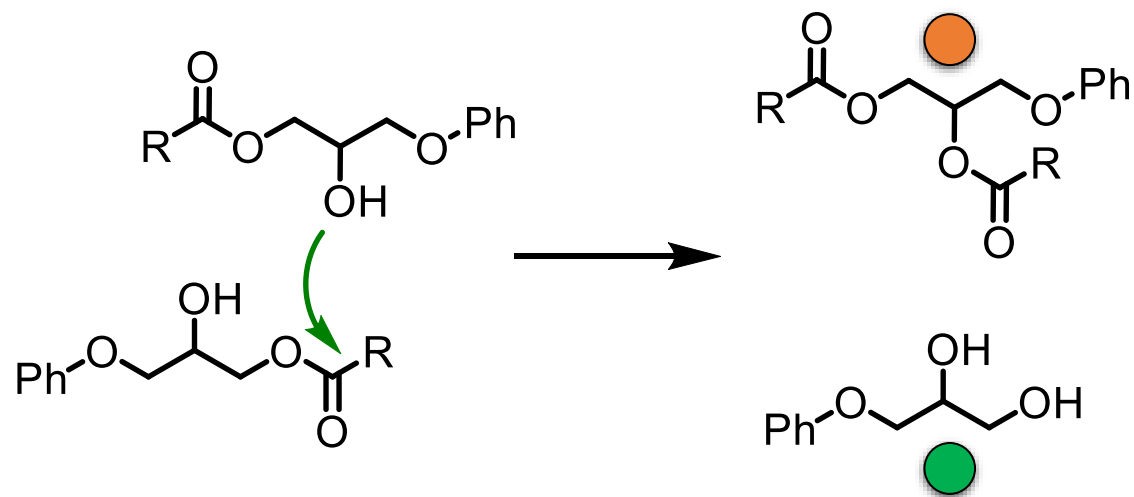
¹H NMR monitoring of epoxy consumption as a function of time



Model molecular reactions: Effect of temperature on epoxy-acid addition



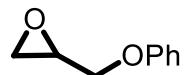
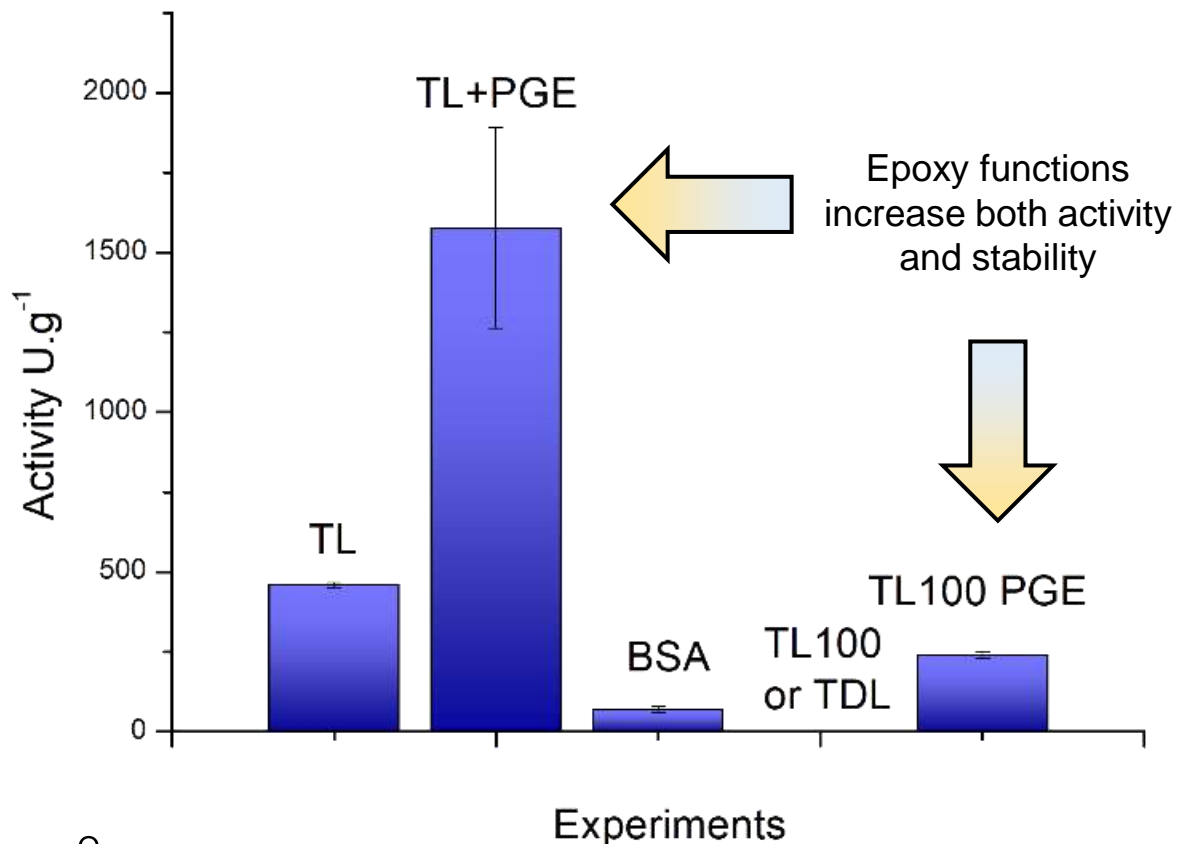
Model molecular reactions: Effect of temperature on transesterification



Transesterification only occurred at temperature below 100°C

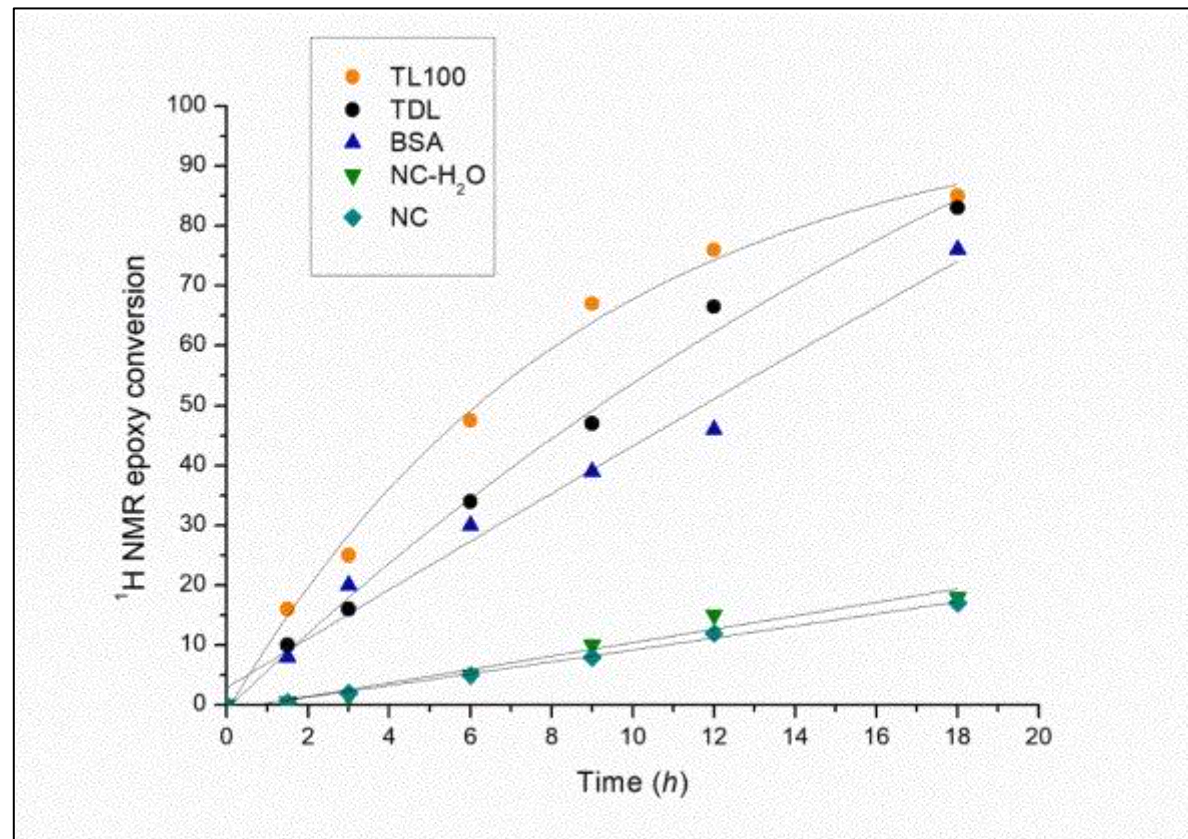
Model molecular reactions: Mechanistic investigation at 100°C

Enzymatic activity (titration of COOH from hydrolysis of triglycerides in emulsion at 50°C)



[PGE]

Kinetic experiments by ¹H NMR at 100°C with various enzymes



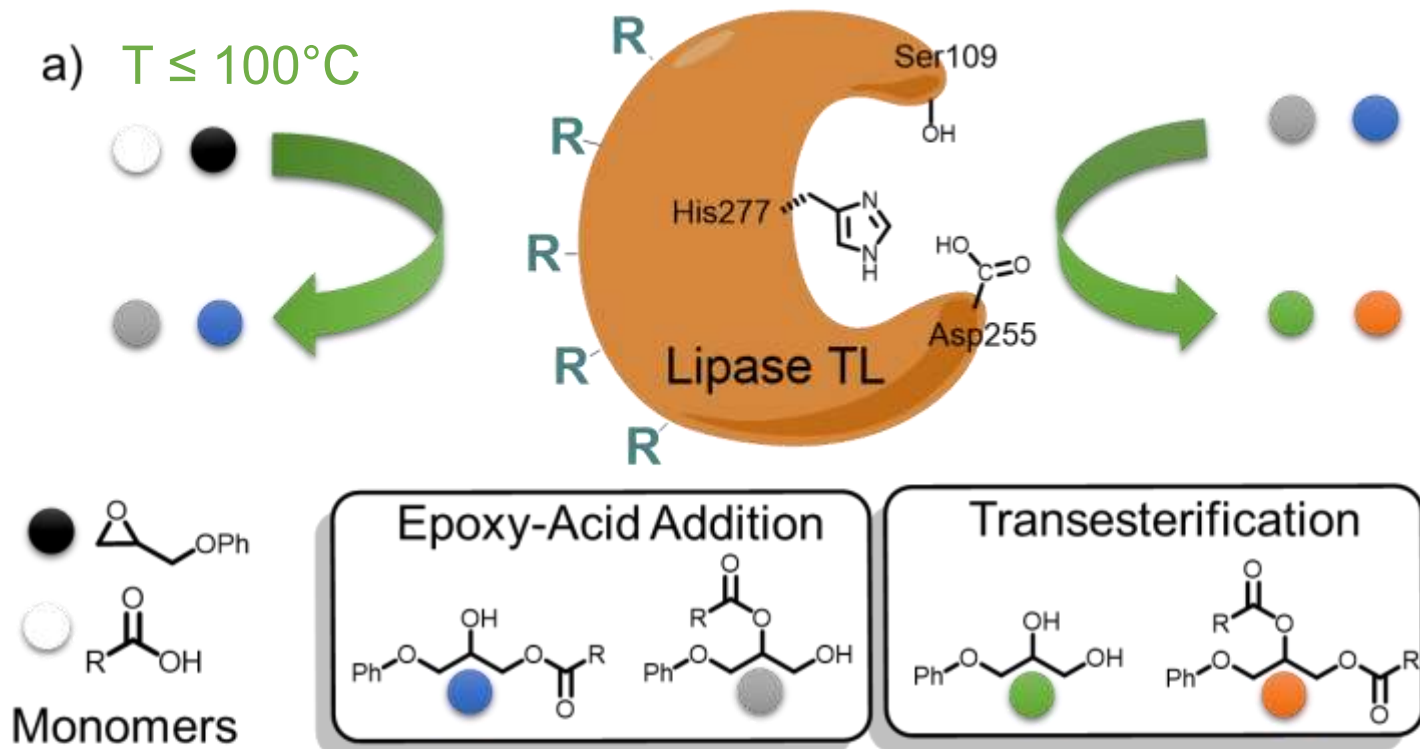
TL = Lipase TL

TDL = Thermally Denaturated Lipase (200°C, 3h)

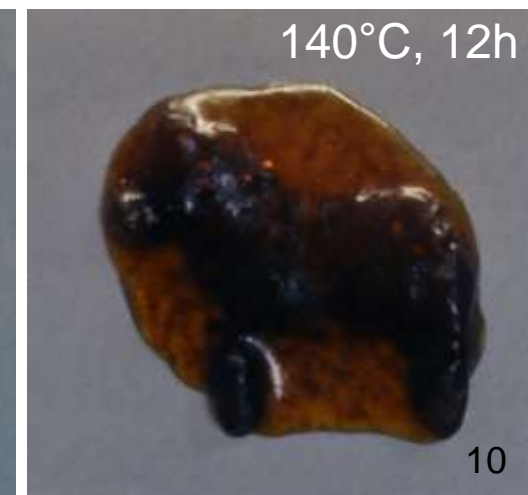
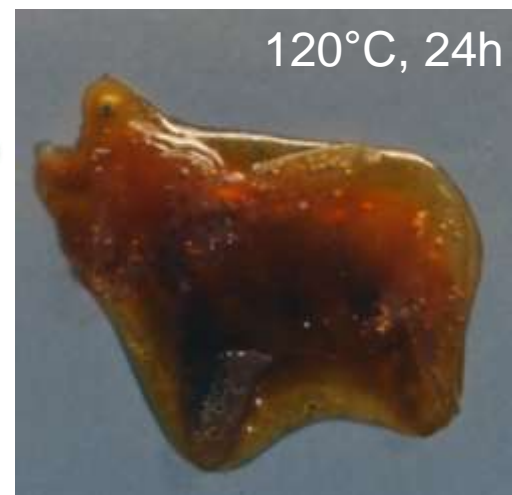
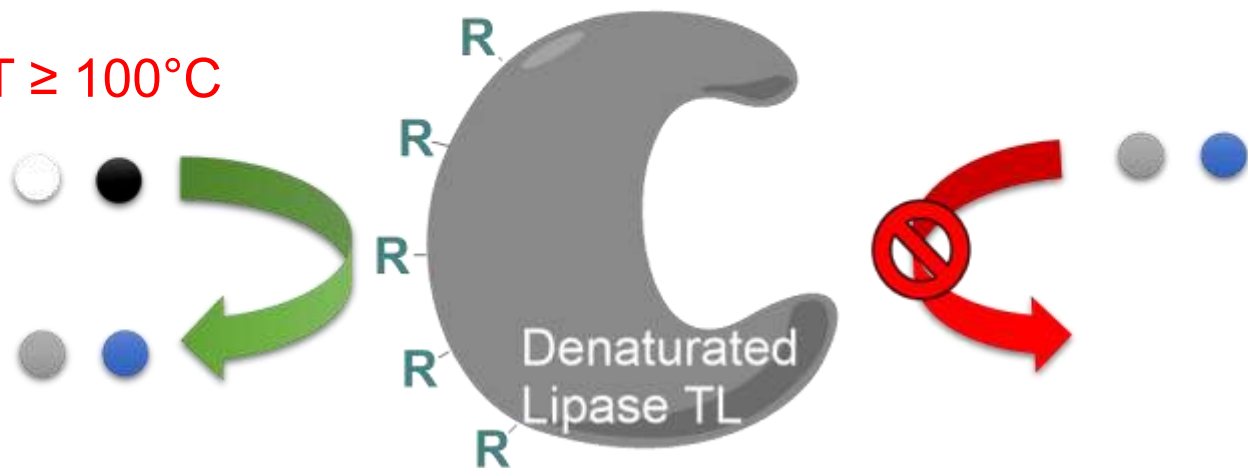
BSA = Bovine Serum Albumin

Model molecular reactions : Proposed mechanism

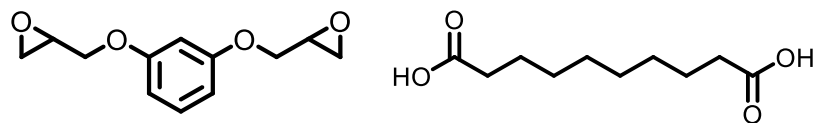
a) $T \leq 100^\circ\text{C}$



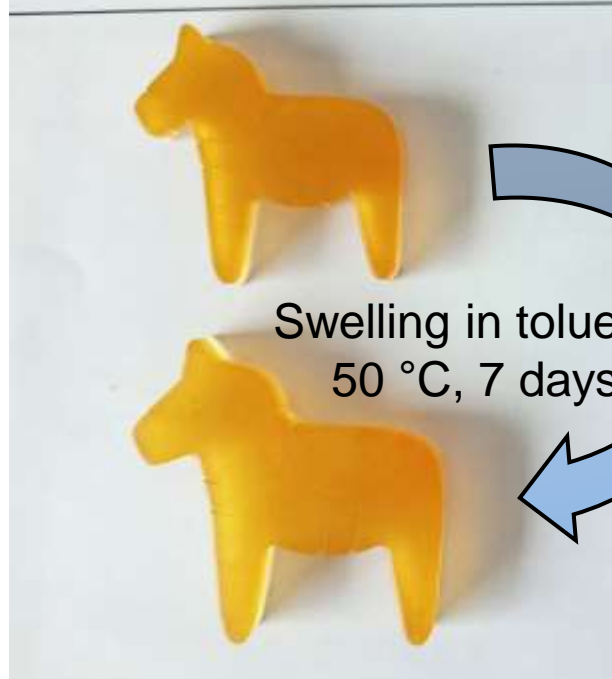
b) $T \geq 100^\circ\text{C}$



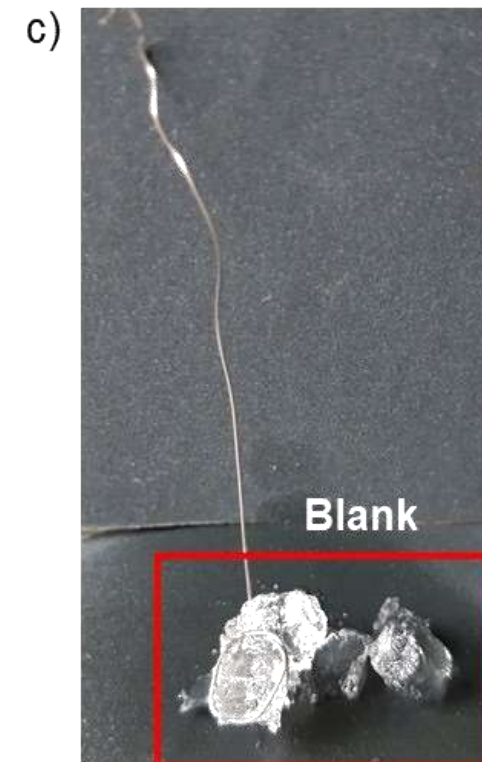
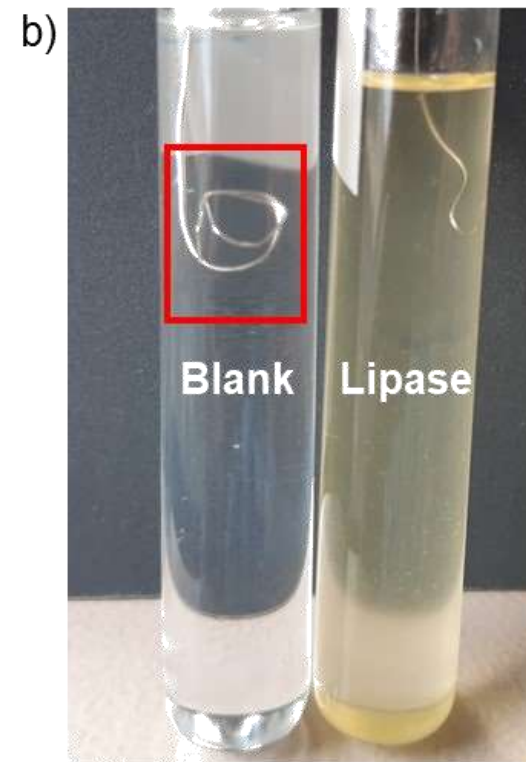
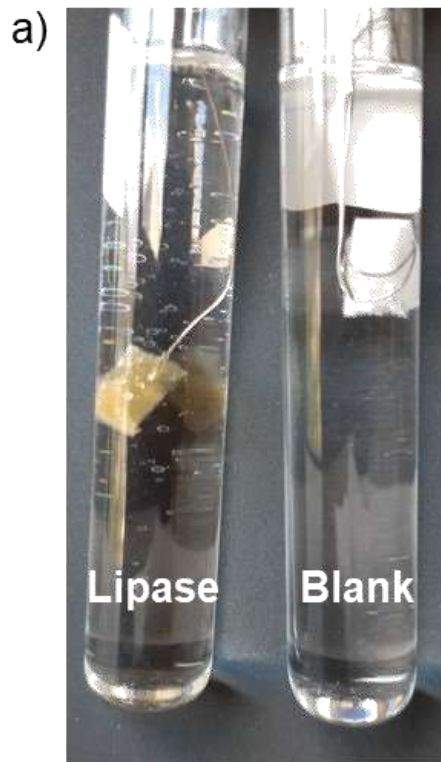
Lipase catalysed polymerisation: A prototype material



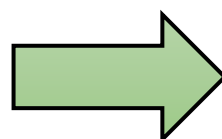
Lipase TL
100°C, 72h



Dissolution test in benzyl alcohol (100°C, 3 days)

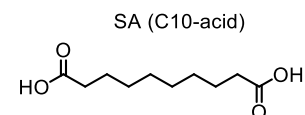
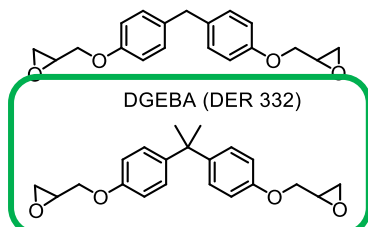
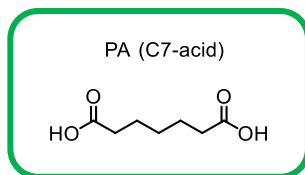
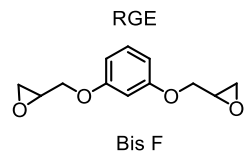
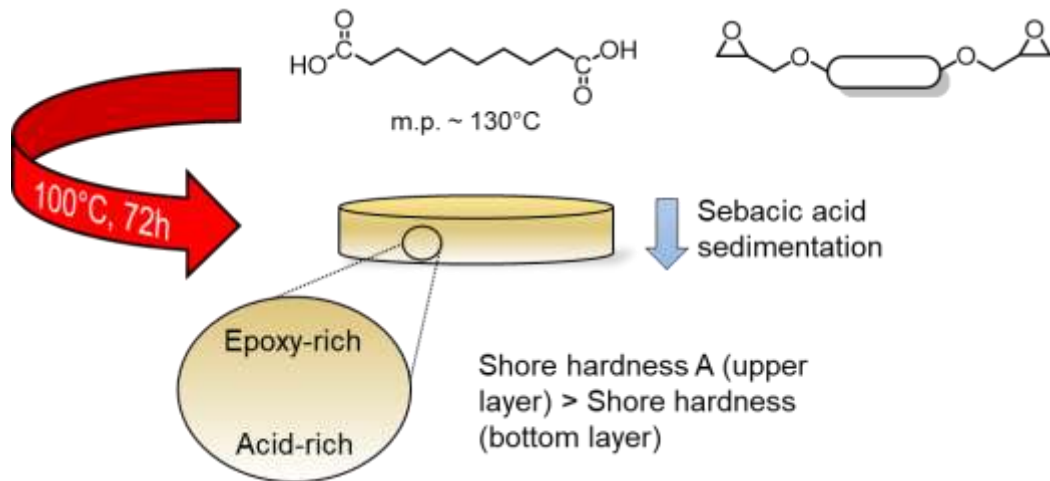


blank = control material synthesized without lipase



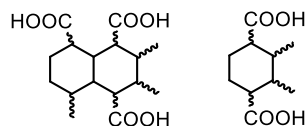
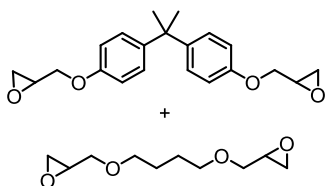
Extracted enzyme from model resin (not crosslinked) : $A \approx 270 \text{ U.g}^{-1}$

Lipase catalysed polymerisation: looking for suitable formulation

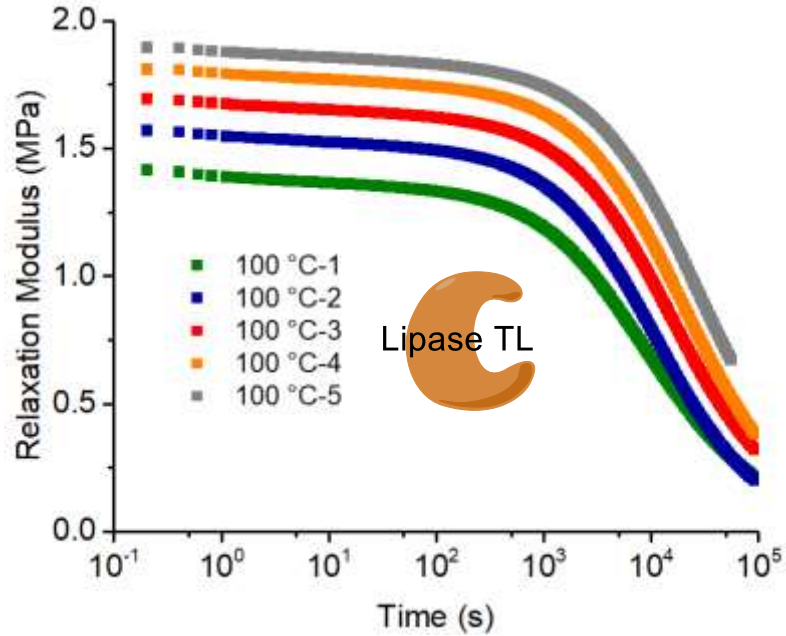


DGEBA + reactive diluent (LY564)

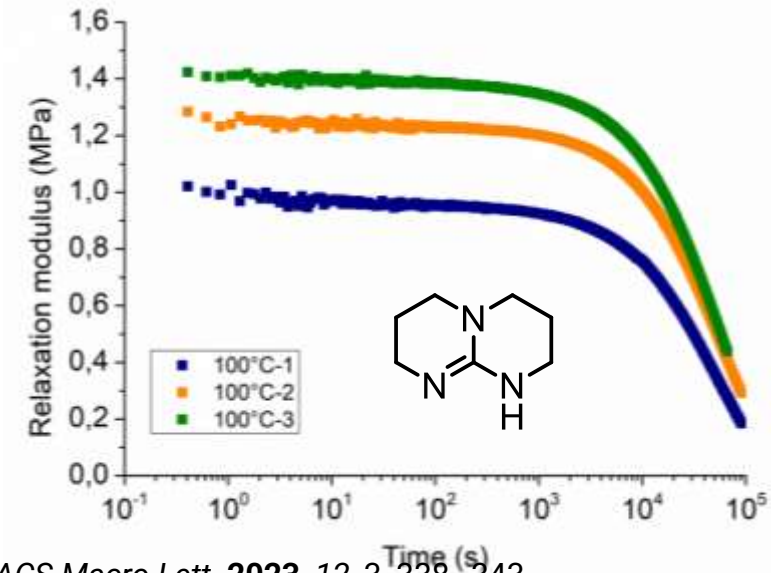
Pripol 1040 (diacid and triacid mixture)



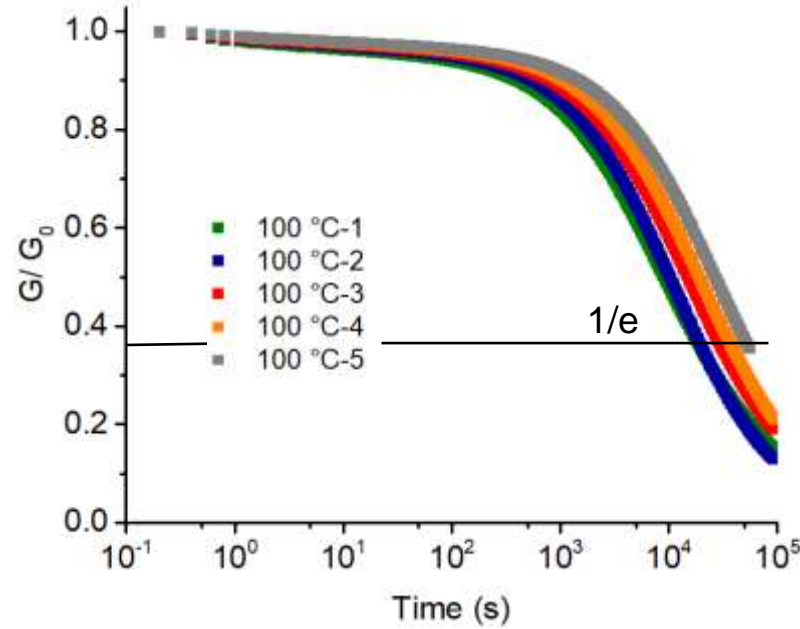
Lipase catalysed polymerisation: vitrimer properties



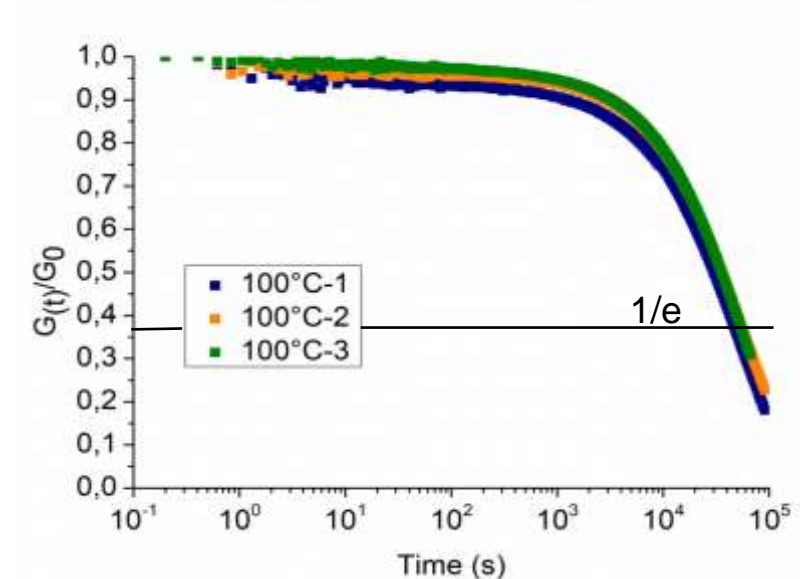
Stress relaxation (torsion, $\gamma=0.5\%$) at 100°C



ACS Macro Lett. 2023, 12, 3, 338–343



Normalised stress relaxation (torsion, $\gamma=0.5\%$) at 100°C

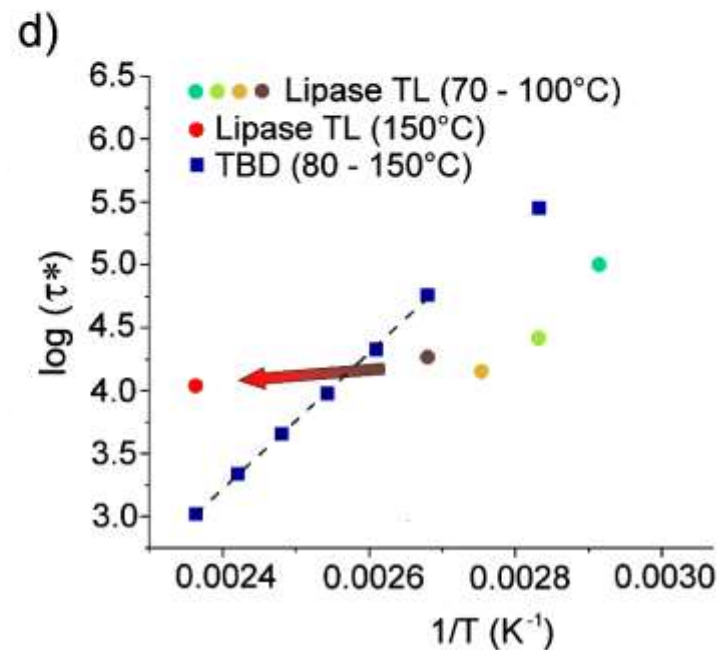
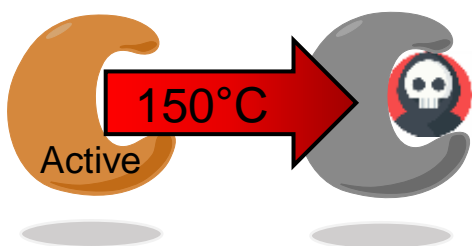
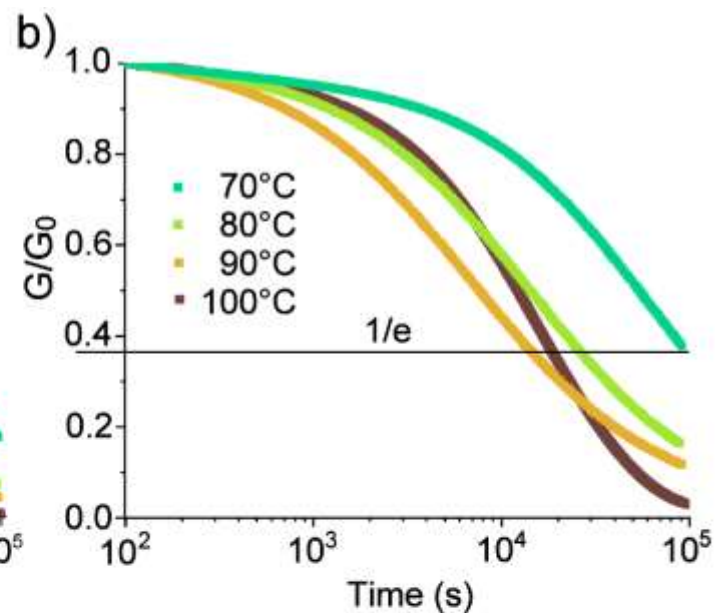
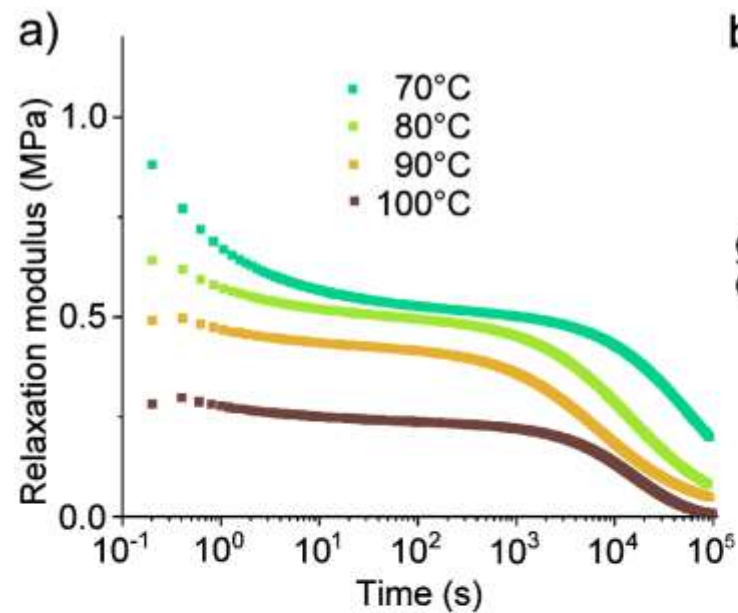
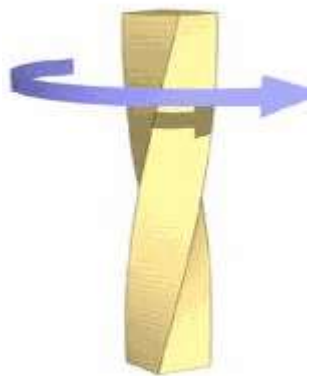


- G_0 increases after each experiments (100°C, 24h)
- $\tau_{(1/e)}$ increases from ~ 4 h to ~ 18 h after 5 days at 100°C

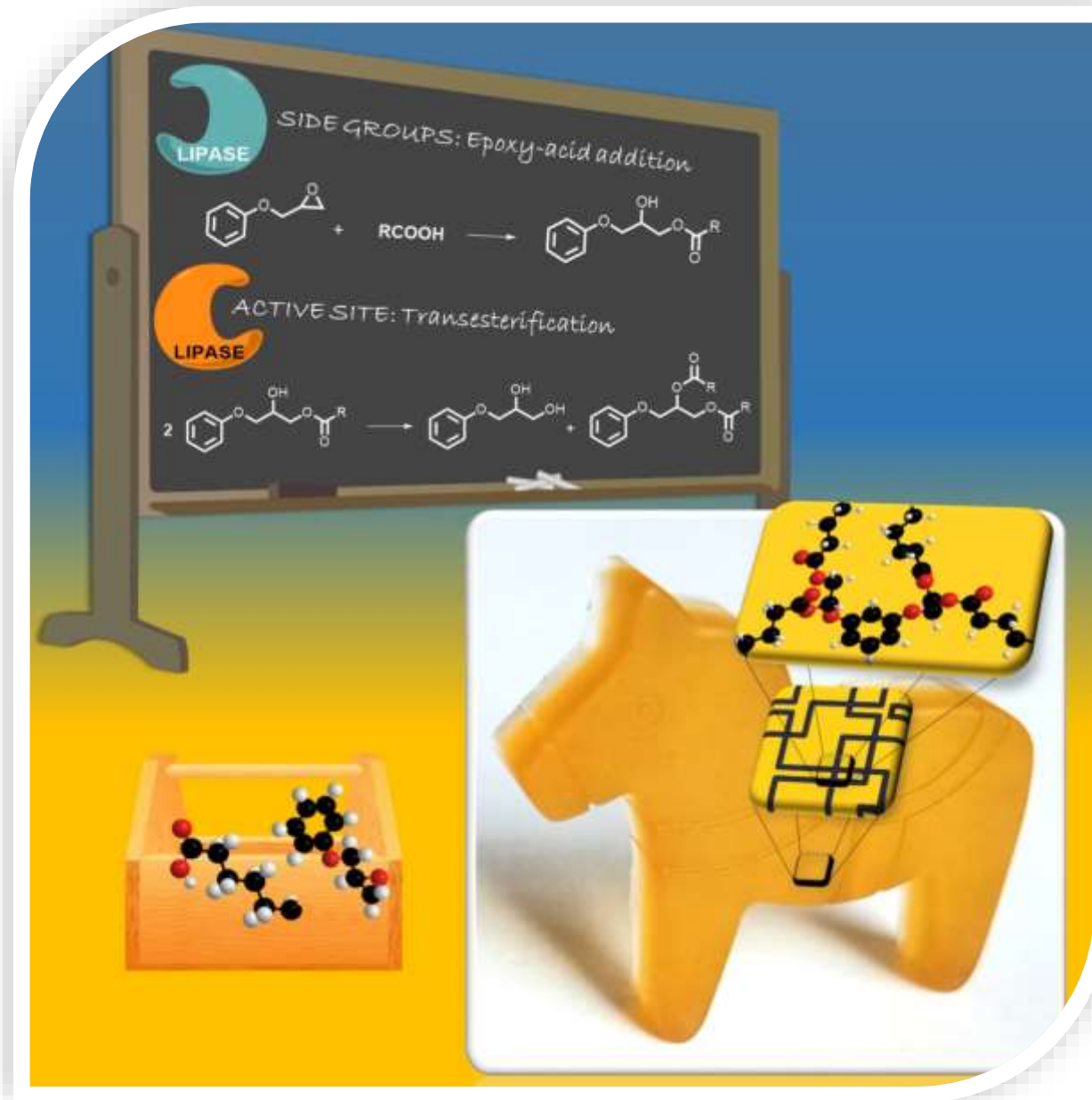
- Similar G_0 increase after each experiments for TBD catalysed material (100°C, 24h)
- $\tau_{(1/e)}$ is stable (~ 18 h)

enzyme denaturation ??

Lipase catalysed polymerisation: vitrimer properties



Conclusion



Molecular Model reactions

- ✓ Suitable conditions for Lipase catalysed epoxy-acid networks
- ✓ Mechanism insight (active site or side groups)

Vitrimer synthesis

- ✓ Suitable formulations for Lipase catalysed epoxy-acid networks
- ✓ Able to relax stress and reprocess multiple times at 100°C
- ✓ Lipase catalyses the exchange reactions

Acknowledgments



Self-HEaling soft RObotics



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Jakob Langenbach
Q.-A. Poutrel
Paolo Edara**

Matthieu Gresil



All of you for your kind attention !